

## Assessment

**Chapter Test A****Chapter: States of Matter**

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas particles are considered to have no definite
- mass.
  - volume.
  - density.
  - energy.
- \_\_\_\_\_ 2. Under which conditions does a real gas behave very much like an ideal gas?
- high temperature and low pressure
  - high temperature and high pressure
  - low temperature and high pressure
  - low temperature and low pressure
- \_\_\_\_\_ 3. If a gas and a liquid are at the same temperature and pressure, diffusion occurs much faster in the gas because
- there are more elastic collisions between the particles in a gas.
  - gases are more compressible.
  - the particles move faster in a gas and there is a greater distance between them.
  - gas molecules are in continuous motion.
- \_\_\_\_\_ 4. If the particles in a sample of matter are attracted to each other but can move past each other easily, the matter is a
- solid.
  - liquid.
  - gas.
  - plasma.
- \_\_\_\_\_ 5. According to the equation for the kinetic energy of a moving object, which gas would have the lowest average velocity, assuming all gases are at the same temperature?
- $O_2$
  - $H_2$
  - $H_2O$
  - $O_3$

# Assessment Chapter Test B States Of Matter

**NARAYAN CHANGDER**



## Assessment Chapter Test B States Of Matter:

**Matter and Energy**, 2005      **Harcourt Science: Life science, units A and B**, 2002      **Glencoe Science**  
McGraw-Hill Staff, 2001-08      *Harcourt Science: Physical science, [grade] 3, units E and F, teacher's ed*, 2000      **States  
and Regions**, 1997      **Sciastmbk Matter and Energy** McGraw Hil, 1999-04      Prentice Hall Physical Science Concepts in  
Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in  
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, 2000      McDougal Littell Science McDougal Littell Incorporated, 2005      *Journal of the House of Representatives of the  
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Studies** S.G. Grant, Bruce A. VanSledright, 2014-03-14 Organized around four commonplaces of education learners and  
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that moves away from traditional high stakes testing towards more meaningful ways of evaluating student learning New chapter that highlights and explains key elements of the Common Core State Standards for English Language Arts and shows how the incorporation of critical ELA instruction into the social studies curriculum can foster more ambitious teaching and learning Real classroom narratives that introduce each chapter and provide in depth access to teaching and learning contexts Practical curriculum and resource suggestions for the social studies classroom End of chapter summaries and annotated teaching resources *Harcourt Science: Physical science, [grade] 4, Units E and F, teacher's ed. [v. 18]. Life science, [grade] 5, Units A and B, teacher's ed ,2000* **Chemical Interactions** ,2005 Harcourt Science: Physical science, [grade] 5, Units E and F, teacher's ed ,2000 Motion, Forces, and Energy Michael J. Padilla,Peter Kahan,2002 **Laws of the State of New York** New York (State), Living in Environment Im Testb Jane Heinze-Fry,Miller,1999-05 **Waves, Sound, and Light** Glencoe/McGraw-Hill,2001-06 Macmillan/McGraw-Hill Science ,2002 An activity based science program

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