

CHEMISTRY PRACTICAL - SALT ANALYSIS

NOTE:

1. Exp. 1 and 2 for your reference, no need to write in your notebook.
2. Exp. 3 to 10 are to be written on the given table.
3. Chemical Equations are to be learned by the students thoroughly.
4. Start each Experiment from new page.

EXPERIMENT – 1 Classification of Anions

Group	Group Reagent	Observation	Inference
A	Dilute H_2SO_4	a) Colourless, odourless gas with brisk effervescence (CO_2) which turns lime water milky. b) Colourless gas with rotten egg like smell (H_2S) which turns lead acetate paper black. c) Colourless gas with smell of burning sulphur (SO_2) which turns acidified dichromate paper green. d) Brown coloured gas (NO_2) which turns ferrous sulphate solution black or brown. e) Colourless gas with vinegar like smell.	CO_3^{2-} (Carbonate) S^{2-} (sulphide) SO_3^{2-} (Sulphite) NO_2^- (Nitrite) CH_3COO^- (Acetate)
B	Conc. H_2SO_4	a) Colourless pungent smelling gas (HCl) which gives white dense fumes with glass rod dipped in NH_4OH . b) Violet coloured vapours (I_2) which turns starch paper blue. c) Reddish brown gas (NO_2) having pungent smell (On adding copper turning, fumes becomes intense) d) Brown colour gas with pungent smell (Br_2) which turns starch paper yellow. e) Colourless, odourless gas with brisk effervescence ($CO + CO_2$) which turns lime water milky and burns on the mouth of test tube with blue flame.	Cl^- (Chloride) I^- (Iodide) NO_3^- (Nitrate) Br^- (Bromide) $C_2O_4^{2-}$ (Oxalate)
C	$BaCl_2$	White ppt. of $BaSO_4$ is formed.	SO_4^{2-} (sulphate)
D	Ammonium molybdate $3(NH_4)_2MoO_4$	Cannary yellow ppt. of phospho ammonium molybdate $(NH_4)_3PO_4 \cdot 12 MoO_3 \cdot 6H_2O$	PO_4^{3-} (phosphate)

EXPERIMENT – 2

Classification of Cations

Group	Group Reagent	Radical	PPt/Smell	Colour
Zero	$NaOH$	NH_4^+	Smell of NH_3	-
I	Dil. HCl	Pb^{2+}	$PbCl_2$	White
II	H_2S gas in acidic Medium	Pb^{2+} As^{3+} Cu^{2+} Cd^{2+}	PbS As_2S_3 CuS CdS	Black Yellow Black Yellow
III	NH_4Cl (s) in presence of NH_4OH	Fe^{3+} Fe^{2+} Al^{3+}	$Fe(OH)_3$ $Fe(OH)_2$ $Al(OH)_3$	Light green Reddish brown Gelatinous white
IV	H_2S gas in basic medium	Ni^{2+}	NiS	Black

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CBSE MCQs Chapterwise For Term I & II, Class 12, Chemistry Career Point Kota, 2021-10-12 Strictly as per the new term wise syllabus for Board Examinations to be held in the academic session 2021 22 for class 12 Multiple Choice Questions based on new typologies introduced by the board Stand Alone MCQs MCQs based on Assertion Reason Case based MCQs Include Questions from CBSE official Question Bank released in April 2021 Answer key with Explanations Sample Paper on the latest pattern of Term 1 exam CBSE Class 12th Chemistry (10 Most Likely Question Papers with Solution) By Career Point Kota Career Point Kota, 2020-07-28 The CBSE has made certain changes in the assessment structure from the session 2019 20 onwards In the new scheme of examination CCE and term system has been replaced with the Internal Assessment Single Annual Exam by CBSE itself Single exam conducted by CBSE will carry 80 marks whereas 20 marks are left to the schools for internal assessment CBSE has issued detailed guidelines on how the internal marks will be divided among different activities From 2019 onwards there will be internal choices in board examinations with increased internal options in the question paper Considering this change now a student has to prepare accordingly for board examinations The new assessment format brought with it excitement as well as anxiety And to help them prepare and excel in their CBSE board examination Career Point Kota has developed a series of 10 Most Likely Question Papers with Solutions The Key Features of Most Likely Question Papers with Solutions Series New OBJECTIVE TYPE question in each paper Syllabus of CBSE 2019 20 Based on the latest CBSE Syllabus Pattern Mind map of each chapter is given to visualize and help acquire a better understanding Important terms facts formulae and quick revision tips are given Covers questions asked in previous year board exams Toppers Answer Sheet as released by CBSE to understand the scoring technique We hope this book will gratify students need for the new CBSE pattern board exam and smoothen their path to success We wish to utilize the opportunity to place on record our special thanks to all the members of the Content Development team for their efforts to create this wonderful book Educart Term 2 Chemistry CBSE Class 12 Objective & Subjective Question Bank 2022 (Exclusively on New Competency Based Education Pattern) EduCart, 2021-12-28 Educart Class 12 Chemistry Question Bank combines remarkable features for Term 2 Board exam preparation Exclusively developed based on Learning Outcomes and Competency based Education Pattern this one book includes Chapter wise theory for learning Solved Questions from NCERT and DIKSHA and Detailed Explanations for concept clearance and Unsolved Self Practice Questions for practice Topper s Answers are also given to depict how to answer Questions according to the CBSE Marking Scheme Solutions Oswaal CBSE Sample Question Papers Class 12 Chemistry (For 2024 Exam) Oswaal Editorial Board, 2023-08-03 Description of the product Fresh Relevant with 2024 CBSE SQP Fully Solved Analyzed Score Boosting Insights with 500 Questions 1000 Concepts Insider Tips Techniques with On Tips Notes Mind Maps Mnemonics Exam Ready to Practice with 10 Highly Probable SQPs with Actual Board Answer sheets Oswaal ISC 10 Sample Question Papers Class 12 Chemistry For 2024

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Table of Contents Class Xii Chemistry Practical Salt Analysis

1. Understanding the eBook Class Xii Chemistry Practical Salt Analysis
 - The Rise of Digital Reading Class Xii Chemistry Practical Salt Analysis
 - Advantages of eBooks Over Traditional Books
2. Identifying Class Xii Chemistry Practical Salt Analysis
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Class Xii Chemistry Practical Salt Analysis
 - User-Friendly Interface
4. Exploring eBook Recommendations from Class Xii Chemistry Practical Salt Analysis
 - Personalized Recommendations
 - Class Xii Chemistry Practical Salt Analysis User Reviews and Ratings
 - Class Xii Chemistry Practical Salt Analysis and Bestseller Lists
5. Accessing Class Xii Chemistry Practical Salt Analysis Free and Paid eBooks

- Class Xii Chemistry Practical Salt Analysis Public Domain eBooks
 - Class Xii Chemistry Practical Salt Analysis eBook Subscription Services
 - Class Xii Chemistry Practical Salt Analysis Budget-Friendly Options
6. Navigating Class Xii Chemistry Practical Salt Analysis eBook Formats
- ePub, PDF, MOBI, and More
 - Class Xii Chemistry Practical Salt Analysis Compatibility with Devices
 - Class Xii Chemistry Practical Salt Analysis Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Class Xii Chemistry Practical Salt Analysis
 - Highlighting and Note-Taking Class Xii Chemistry Practical Salt Analysis
 - Interactive Elements Class Xii Chemistry Practical Salt Analysis
8. Staying Engaged with Class Xii Chemistry Practical Salt Analysis
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Class Xii Chemistry Practical Salt Analysis
9. Balancing eBooks and Physical Books Class Xii Chemistry Practical Salt Analysis
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Class Xii Chemistry Practical Salt Analysis
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Class Xii Chemistry Practical Salt Analysis
- Setting Reading Goals Class Xii Chemistry Practical Salt Analysis
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Class Xii Chemistry Practical Salt Analysis
- Fact-Checking eBook Content of Class Xii Chemistry Practical Salt Analysis
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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