

Stoichiometry Worksheet # 1

1) Glucose is used as a source of energy by the human body. The overall reaction in the body is



Calculate the number of moles of oxygen needed to oxidize 12.5 g of $\text{C}_6\text{H}_{12}\text{O}_6$.

2) Ammonia is synthesized from hydrogen and nitrogen according to the following equation.



If an excess of nitrogen is reacted with 3.41 moles of hydrogen gas, how many grams of ammonia (NH_3) can be produced?

3) Assume that in the decomposition of potassium chlorate, KClO_3 ,



80.5 moles of O_2 form. How many grams of potassium chlorite, the other product would be formed?

4) In a single displacement reaction,



9.23 moles of aluminum react with excess hydrochloric acid. How many moles of hydrogen will be produced?

5) The compound "Cisplatin" $\text{PtCl}_2(\text{NH}_3)_2$, has been found to be effective in treating some types of cancer. It can be synthesized using the following reaction



A. How many moles of "cisplatin" can be produced from 2.50 g K_2PtCl_4 ?

B. How many moles of NH_3 would be needed if 3.8 moles of K_2PtCl_4 were present?

Chapter 12 Stoichiometry Answer

M Carnoy



Chapter 12 Stoichiometry Answer:

This book delves into Chapter 12 Stoichiometry Answer. Chapter 12 Stoichiometry Answer is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chapter 12 Stoichiometry Answer, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Chapter 12 Stoichiometry Answer
- Chapter 2: Essential Elements of Chapter 12 Stoichiometry Answer
- Chapter 3: Chapter 12 Stoichiometry Answer in Everyday Life
- Chapter 4: Chapter 12 Stoichiometry Answer in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Chapter 12 Stoichiometry Answer. The first chapter will explore what Chapter 12 Stoichiometry Answer is, why Chapter 12 Stoichiometry Answer is vital, and how to effectively learn about Chapter 12 Stoichiometry Answer.

3. In chapter 2, this book will delve into the foundational concepts of Chapter 12 Stoichiometry Answer. The second chapter will elucidate the essential principles that need to be understood to grasp Chapter 12 Stoichiometry Answer in its entirety.

4. In chapter 3, this book will examine the practical applications of Chapter 12 Stoichiometry Answer in daily life. This chapter will showcase real-world examples of how Chapter 12 Stoichiometry Answer can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Chapter 12 Stoichiometry Answer in specific contexts. The fourth chapter will explore how Chapter 12 Stoichiometry Answer is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Chapter 12 Stoichiometry Answer. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Chapter 12 Stoichiometry Answer.

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