

# Stoichiometry

## Section 11.1 What is stoichiometry?

In your textbook, read about stoichiometry and the balanced equation.

For each statement below, write *true* or *false*.

- \_\_\_\_\_ 1. The study of the quantitative relationships between the amounts of reactants used and the amounts of products formed by a chemical reaction is called stoichiometry.
- \_\_\_\_\_ 2. Stoichiometry is based on the law of conservation of mass.
- \_\_\_\_\_ 3. In any chemical reaction, the mass of the products is less than the mass of the reactants.
- \_\_\_\_\_ 4. The coefficients in a chemical equation represent not only the number of individual particles but also the number of moles of particles.
- \_\_\_\_\_ 5. The mass of each reactant and product is related to its coefficient in the balanced chemical equation for the reaction by its molar mass.

Complete the table below, using information represented in the chemical equation for the combustion of methanol, an alcohol.

methanol + oxygen  $\rightarrow$  carbon dioxide + water



| Substance         | Molar Mass (g/mol) | Number of Molecules | Number of Moles (mol) | Mass (g) |
|-------------------|--------------------|---------------------|-----------------------|----------|
| 6. Methanol       | 32.05              |                     |                       |          |
| 7. Oxygen gas     | 32.00              |                     |                       |          |
| 8. Carbon dioxide | 44.01              |                     |                       |          |
| 9. Water          | 18.02              |                     |                       |          |

10. What are the reactants? \_\_\_\_\_
11. What are the products? \_\_\_\_\_
12. What is the total mass of the reactants? \_\_\_\_\_
13. What is the total mass of the products? \_\_\_\_\_
14. How do the total masses of the reactants and products compare? \_\_\_\_\_

# Chapter 11 Study For Content Mastery The Mole

**Michal Rosen-Zvi**



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serpentinite is composed of one or more serpentine group minerals formed by hydration and low temperature metamorphic transformation of ultramafic rocks from platinum nickel chromium deposits 2017

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serpentine is a large mineral group with hydrous magnesium iron silicate that forms from the serpentinization of rocks like peridotite and olivine as a silicate crystal it forms many varieties of green brown blue red black white pink orange and yellow colors with a greasy silky soapy or flaky texture

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serpentinite is a rock composed predominantly of one or more serpentine group minerals the name originating from the similarity of the texture of the rock to that of the skin of a snake serpentinite has been called serpentine or serpentine rock particularly in older geological texts and in wider cultural settings

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serpentine any of a group of hydrous magnesium rich silicate minerals the composition of these common rock forming minerals approximates  $\text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4$  serpentine generally occurs in three polymorphs chrysotile a fibrous variety used as asbestos antigorite a variety occurring in either

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serpentine is a group of minerals that are usually green in color it is used as gemstone architectural stone carving material and source of asbestos serpentinite is a rock composed mainly of serpentine minerals

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