

# OFFICIAL WORKSHOP

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you  $23 \text{ Fe} + \text{Co} \rightarrow \text{Co} + 2 \text{Fe}$   
oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$  reduction  $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$   $24 \text{ 3 Ag} + \text{Ni} \rightarrow \text{Ni} + 3 \text{ Ag}$  oxidation reduction  $25 \text{ Cu} + \text{Pb} \rightarrow \text{Pb} + \text{Cu}$  oxidation reduction  $26 \text{ O}_2$

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 $5\text{CO} + \text{I}_2 \rightarrow 5\text{CO}_2 + \text{I}_2$   $\text{O}_5 + \text{I}_2 \rightarrow \text{O}_2 + \text{I}_2$   $5\text{CO} + \text{I}_2 \rightarrow 5\text{CO}_2 + \text{I}_2$

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