

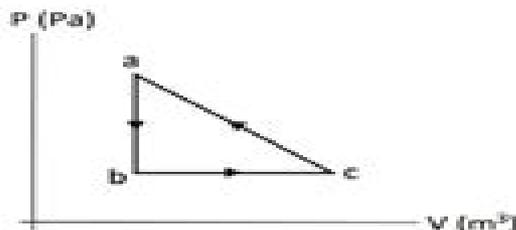
First Law of Thermodynamics

Grade

Name _____ Date _____

Use the following P-V diagram to answer questions about this thermodynamic cycle consisting of three processes labeled ab , bc , and ca .

$$P_a = 3.00 \times 10^5 \text{ Pa}, P_b = 1.00 \times 10^5 \text{ Pa}, V_a = 3 \times 10^{-3} \text{ m}^3, V_c = 9 \times 10^{-3} \text{ m}^3.$$



a) Identify the type of thermodynamic processes for each part of the cycle (ab , bc , ca) as:

- | | | | |
|--------|----------------|------------------|----------------|
| ab : | i) adiabatic | ii) isobaric | iii) isochoric |
| | iv) isothermal | v) none of these | |
| bc : | i) adiabatic | ii) isobaric | iii) isochoric |
| | iv) isothermal | v) none of these | |
| ca : | i) adiabatic | ii) isobaric | iii) isochoric |
| | iv) isothermal | v) none of these | |

b) For the process ab , determine (i) the work W_{ab} done by the gas and (ii) the change in internal energy of the system for a heat loss of 900 J. [Hints: for (i) recall that W is the area under the curve; for (ii) use the 1st Law.]

c) For the process bc , determine (i) the work W_{bc} done by the gas and (ii) the change in internal energy of the system for a heat gain of 1250 J.

d) For the process ca , determine (i) the work W_{ca} done by the gas and (ii) the change in internal energy of the system for a heat gain of 150 J.

e) Determine the total work W_{tot} done by the gas in the cycle. Compare this to the area of the triangle. Note that both are negative. Also note that the total change in internal energy is zero, as it must be when the system returns to the same state.

Thermodynamic Questions And Answers

Chaejoon Cheong



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Fundamentals of Engineering Thermodynamics R. Yadav ,2023-05-10 Thermodynamics deals with energy interactions between material bodies It is the science of 3E s namely Energy Entropy and Equilibrium The applications of its laws and principles are found in all fields of energy technology notably in steam gas and nuclear power plants internal combustion engines gas turbines jet propulsion refrigeration air conditioning compressors gas dynamics and direct energy conversion Starting with the basic concept the book discusses the important topics such as basic concepts heat and work energy ideal and real gases zeroth first and second laws of thermodynamics entropy and third law available energy and exergy gas power cycles vapour power cycles general thermodynamic relations refrigeration cycles psychrometry non reactive mixtures reactive mixture chemical equilibrium direct energy conversion compressible flows and heat transfer The book is an essential text for BE B Tech for Mechanical Engineering students UPSC and GATE examinations

Thermodynamics and Thermal Engineering J.Selwin Rajadurai,2003 Thermodynamics And Thermal Engineering A Core Text In SI Units Meets The Complete Requirements Of The Students Of Mechanical Engineering In All Universities Ultimately It Aims At Aiding The Students Genuinely Understand The Basic Principles Of Thermodynamics And Apply Those Concepts To Practical Problems Confidently It Provides A Clear And Detailed Exposition Of Basic Principles Of Thermodynamics Concepts Like Enthalpy Entropy Reversibility Availability Are Presented In Depth And In A Simple Manner Important Applications Of Thermodynamics Like Various Engineering Cycles And Processes Are Explained In Detail Introduction To Latest Topics Are Enclosed At The End Each Topic Is Further Supplemented With Solved Problems Including Problems From Gate Ies Exams Objective Questions Along With Answers Review Questions And Exercise Problems Alongwith Answers For An Indepth Understanding Of The Subject

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