

**CLASS** : **X (2024 25)**  
**SUBJECT** : **PHYSICAL SCIENCES**  
**CHAPTER** : **5. CARBON AND ITS COMPOUNDS**

### **CARBON AND ITS COMPOUNDS - REVISION WORKSHEET**

1. What would be the electron dot structure of carbon dioxide which has the formula of  $\text{CO}_2$ ?
2. How many structural isomers can you draw for pentane? Draw them.
3. What are two properties of carbon which lead to the huge number of carbon compounds we see around us?
4. Draw the structure for the following compounds :
  - (i) Ethanoic acid
  - (ii) Bromopentane
  - (iii) Butane
  - (iv) Hexanal
5. How would you name the following compounds?
6. Draw the electron dot structure for
  - (a) Ethanoic acid
  - (b)  $\text{H}_2\text{S}$
  - (c) Propanone
  - (d)  $\text{F}_2$
7. What is a homologous series? Explain with an example.
8. How can ethanol and Ethanoic acid be differentiated on the basis of their physical and chemical properties?
9. Why does micelle formation take place when soap is added to water? Will a micelle be formed in other solvents such as ethanol also?
10. Why are carbon and its compounds used as fuels for most applications?
11. Explain the formation of scum when hard water is treated with soap.
12. What change will you observe if you test soap with litmus paper (red and blue)?
13. What is hydrogenation? What is its industrial application?
14. Give a test that can be used to differentiate chemically between butter and cooking oil?
15. Explain in mechanism of the cleaning action of soap.
16. Write the electron dot structure of ethane molecule,  $\text{C}_2\text{H}_6$ .
17. Name the functional group present in each of the following compounds:
  - (i)  $\text{HCOOH}$
  - (ii)  $\text{C}_2\text{H}_5\text{CHO}$
18. Write the name and structure of an alcohol with three carbon atoms in its molecule.
19. Write the name and structure of an alcohol with four carbon atoms in its molecule.
20. Write the name and structure of an aldehyde with four carbon atoms in its molecule.
21. Name the main products formed when

# Section 1 Carbon Compounds Answers

**A Loxley**



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