

MOLE TO MOLE STOICHIOMETRY WORKSHEET

Name : _____

Date : _____

1. Using the equation



a. Find the number of moles of Al used to produce 9.9 moles of H_2 .

b. Find the number of moles of HCl used to produce 1.11 moles of AlCl_3 .

c. Find the number of moles of AlCl_3 produced along with 0.0757 moles of H_2 .

2. The reaction between lithium and fluorine produces lithium fluoride. How many moles of lithium fluoride can be produced with 2.17 mol of fluorine?



3. In the reaction of copper (II) oxide breaking down to form copper and oxygen, how much copper is produced if 0.663 moles of oxygen is given off?



4. Fermentation is a complex chemical process of making wine by converting glucose into ethanol and carbon dioxide:



If 17.5 moles of ethanol were produced, how many moles of glucose were there initially?

5. Consider the reaction:



Calculate the moles of HCl needed to react completely with 8.25 moles of zinc.

Moles And Stoichiometry Practice Problems Answers

Daniela Niemeyer



Moles And Stoichiometry Practice Problems Answers:

Adopting the Tune of Appearance: An Psychological Symphony within **Moles And Stoichiometry Practice Problems Answers**

In some sort of used by displays and the ceaseless chatter of instantaneous communication, the melodic splendor and emotional symphony produced by the written word usually diminish in to the back ground, eclipsed by the constant noise and distractions that permeate our lives. But, nestled within the pages of **Moles And Stoichiometry Practice Problems Answers** an enchanting literary prize brimming with raw thoughts, lies an immersive symphony waiting to be embraced. Constructed by a masterful composer of language, that interesting masterpiece conducts visitors on a mental journey, well unraveling the hidden tunes and profound impact resonating within each carefully crafted phrase. Within the depths of the emotional assessment, we can explore the book is main harmonies, analyze their enthralling publishing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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