

## Chapter Test B

### Chapter Chemical Equations and Reactions

Write the full name (including units) for the term at the end of the sentence. Please show your work. Please write answers without answers with symbols.

- The gas volume of a diatomic molecule is  $2.0 \times 10^3$  L. How many molecules of gas are there?
  - gas
  - gas volume
  - number of molecules
  - liters
- The volume of a gas is  $2.0 \times 10^3$  L. How many molecules of gas are there?
  - number of molecules
  - number of molecules
  - number of molecules
  - number of molecules
- The volume of a gas is  $2.0 \times 10^3$  L. How many molecules of gas are there?
  - number of molecules
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  - number of molecules
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  - number of molecules
  - number of molecules
  - number of molecules
  - number of molecules

# Modern Chemistry Chapter 7 Test B Answers

**AN Whitehead**



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