

Periodic Trends Notes

- The periodic table organizes all known elements according to their number of protons.
- The unique shape of the table reveals the ways in which the electrons are organized around the atom in energy levels.

Periodic Table of Elements

| | | | | | | | | | | | | | | | | | | |
|----|----|----|-----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|
| 1A | | | | | | | | | | | | | | | | | 8A | |
| 1 | H | | | | | | | | | | | | | | | | | He |
| 2 | Li | Be | | | | | | | | | | | B | C | N | O | F | Ne |
| 3 | Na | Mg | 3B | 4B | 5B | 6B | 7B | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |
| 4 | K | Ca | Sc | Ti | V | Cr | Mn | Fe | Co | Ni | Cu | Zn | Ga | Ge | As | Se | Br | Kr |
| 5 | Rb | Sr | Y | Zr | Nb | Mo | | Ru | Rh | Pd | Ag | Cd | In | Sn | Sb | Te | I | Xe |
| 6 | Cs | Ba | *La | Hf | Ta | W | Re | Os | Ir | Pt | Au | Hg | Tl | Pb | Bi | Po | At | Rn |
| 7 | Fr | Ra | *Ac | | | | | | | | | | | | | | | |

• Lanthanide Series

| | | | | | | | | | | | | | |
|----|----|----|----|----|----|----|----|----|----|----|----|----|----|
| 58 | 59 | 60 | 61 | 62 | 63 | 64 | 65 | 66 | 67 | 68 | 69 | 70 | 71 |
| Ce | Pr | Nd | | Sm | Eu | Gd | Tb | Dy | Ho | Er | Tm | Yb | Lu |

• Actinide Series

| | | | | | | | | | | | | | |
|----|----|----|----|----|----|----|----|----|----|-----|-----|-----|-----|
| 90 | 91 | 92 | 93 | 94 | 95 | 96 | 97 | 98 | 99 | 100 | 101 | 102 | 103 |
| Th | Pa | U | | | | | | | | | | | |

Legend - click to find out more...

| | | | |
|---------------|---------------------|-------------------|----------------|
| H - gas | Li - solid | Br - liquid | Tc - synthetic |
| Non-Metals | Transition Metals | Rare Earth Metals | Halogens |
| Alkali Metals | Alkali Earth Metals | Other Metals | Inert Elements |

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