

Chemistry of Life

DOODLE NOTES

Chemistry of Life

BASIC CHEMISTRY REVIEW

atoms and life

All living things are made of atoms.
An atom is the smallest part of an element. Many different types of atoms exist and they are arranged in the periodic table of elements. Atoms are made up of subatomic particles.

Label the structure of this oxygen atom:

• Note the charge and mass of each subatomic particle.

Charge	Mass
Electron	-1
Neutron	1
Proton	1

• How many shells are in the first shell? 2
• How many shells are in the outer shell? 2
• What do we call electrons in the outer shell? Valence electrons

elements

An element is a substance that is made of one type of atom only. They cannot be broken down into simpler substances.

compounds

A compound is made of two or more types of atoms that are chemically bonded. They can be broken down chemically. There are both compounds and mixtures.

mixtures

A mixture is a mix of two or more types of atoms that are NOT chemically bonded. They can be separated physically. There are solutions, suspensions, and emulsions.

ionic bonding

Electron transfer from a metal to a non-metal.

ionic compound

Na⁺ Cl⁻

NaCl

ionic bonding

Mg²⁺ O²⁻

MgO

functional groups

These particular groups of atoms are attached to the carbon chain to give the properties that determine them. They are called functional groups.

Alcohol

-OH

Carbonyl

C=O

Carboxyl

-COOH

Hydroxyl

-OH

Phosphate

-PO₄

Amino

-NH₂

Acidic

-COOH



MULTIPLE VERSIONS & EXAMPLE

Biology The Chemistry Of Life Answer Key

JA Banks



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