

BIOCHEMISTRY TEST – PRACTICE QUESTIONS (Answers on last page)

- The atomic # for Fe (iron) is 26. How many protons in Fe^{2+} ?
(A) 22 (B) 24 (C) 26 (D) 28 (E) 30
- Two molecules that are **isomers**:
(A) must contain the same functional groups
(B) often differ in the number of unsaturated bonds they possess
(C) have the same molecular formulas
(D) often have different masses
- The maximum number of **electrons** which may occupy any given energy level in an atom is:
(A) the Group number (D) $2n^2$
(B) the valence of that atom (E) the octet rule
(C) n^2
- In which of the following kinds of organic compounds does a **carbon atom bond ONLY to hydrogen and other carbon atoms**?
(A) carbohydrates (D) nucleic acids
(B) fatty acids (E) hydrocarbons
(C) lipids
- Glycogen is best defined as a:
(A) storage form of lipids (D) polymer of glucose
(B) storage form of glycerol (E) storage form of proteins
(C) polymer of lactose
- Most **biologically important atoms** attempt to possess _____ electrons in their outermost energy level.
(A) n^2 (B) 2 (C) 10 (D) 4 (E) 6
- The number of **valence** electrons in an atom with atomic number 13 is:
(A) 1 (B) 3 (C) 5 (D) 8 (E) 13
- Which molecule does **NOT** have a **polar covalent** bond?
(A) H_2O (B) H-Br (C) H-H (D) H-F (E) H-I
- What chemical family does $\text{H}_2\text{C}=\text{C}=\text{CH}_2$ belong to?
(A) alkane (D) ether
(B) alkene (E) ester
(C) alkyne
- In the **induced-fit model** of enzyme action, a _____ must bind to the enzyme's _____ for the enzyme to perform its function.
(A) catalyst; activation energy (D) product; catalytic site
(B) product; active site (E) water molecule; allosteric site
(C) substrate; active site
- Which one of the following applies to a **basic solution**?
(A) high OH^- (D) $\text{H}^+ = \text{OH}^-$
(B) low OH^- (E) both B and C are correct
(C) high H^+
- Which one of the four classes of biologically important molecules does **CELLULOSE** belong to?
(A) carbohydrates (C) proteins
(B) nucleic acids (D) lipids
- Amino acids** are joined during a(n) _____ reaction and a(n) _____ bond/linkage is formed.
(A) hydrolysis; peptide (C) condensation; amide
(B) neutralization; ether (D) oxidation; ester
- With respect to proteins, the **primary structure** refers to the:
(A) helix or pleated sheet pattern of a polypeptide
(B) complex 3-dimensional folding pattern of a polypeptide
(C) interaction of more than one polypeptide chain
(D) ordered sequence of amino acids in a polypeptide

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**Frederick A. Bettelheim, William M.
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element with the highest electron affinity is fluorine

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web periodic trends straw lab objective to create a visual representation of the following periodic trends atomic size

ionization energy and electronegativity atomic radius the size of an atom measured in either nanometers nm or angstroms Å

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web you might need periodic table the first two ionization energies for beryllium are shown below be g be g e be g be 2 g e i 1 900 kJ mol i 2 1757 kJ mol which of the following identifies the most probable value for the third ionization energy for be and provides the best justification

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web caragabriel preview chemistry test oct 7 11 43 terms misterlominyo preview terms in this set 25 each element has 3 numbers listed under it which value represents the atomic radius the first number below the diagram what are the units for the atomic radius pm picometers 10 negative twelve meters

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