

- C. repulsion of molecules
- D. attraction of molecules
- E. surface tension of molecules.

**Answer: B**

**16. According to kinetic theory of gases, the absolute zero temperature is attained when**

- A. volume of the gas is zero
- B. pressure of the gas is zero
- C. kinetic energy of the molecules is zero
- D. specific heat of gas is zero
- E. mass is zero.

**Answer: C**

**17. Kinetic theory of gases assumes that the collisions between the molecules are**

- A. perfectly elastic
- B. perfectly inelastic
- C. partly elastic
- D. partly inelastic
- E. partly elastic and partly inelastic.

**Answer: A**

**18. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to**

- A. E/3
- B. E/2
- C. 2E/3

**Answer: A**

**19. The**

**Answer: A**

**20. The kinetic energy of the molecules in terms of absolute temperature T is proportional to**

- A. T
- B. T<sup>2</sup>
- C. 1/T
- D. 1/T<sup>2</sup>
- E. T<sup>3/2</sup>

**Answer: A**

**21. The temperature of a gas is directly proportional to**

- A. average kinetic energy
- B. average potential energy
- C. average total energy
- D. average momentum
- E. average mass of gas molecules

**Answer: A**

**22. The absolute zero temperature is defined as**

- A. the temperature at which the volume of a gas becomes zero
- B. the temperature at which the pressure of a gas becomes zero
- C. the temperature at which the kinetic energy of the molecules becomes zero

# Thermodynamics Multiple Choice Questions Answers

**Xiaolong Qi**



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