

SEVENTH EDITION

PERRY'S CHEMICAL ENGINEERS' HANDBOOK

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Robert H. Perry
Don W. Green

Perry Chemical Engineering Handbook 7th Edition

KJ Lindholm-Leary



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and 5 are dedicated to adsorption ion exchange and catalysis respectively The basic principles are presented including kinetics equilibrium mass heat transfer phenomena as well as the analytical solutions of the reactor models presented in Chapter 3 In the sixth chapter the subject of scale up is approached The two Annexes at the end of the book contain physical properties of substances of environmental interest as well as unit conversion tables Finally nearly all the examples contained are based on real experimental data found in literature with environmental interest Most of the examples consider all aspects of operation design kinetics hydraulics and mass transfer Provides basic knowledge of major environmental problems and connects them to chemical engineering

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confidence Drawing on his award winning courses at Penn State Dr Themis Matsoukas focuses on why as well as how He offers extensive imagery to help students conceptualize the equations illuminating thermodynamics with more than 100 figures as well as 190 examples from within and beyond chemical engineering Part I clearly introduces the laws of thermodynamics with applications to pure fluids Part II extends thermodynamics to mixtures emphasizing phase and chemical equilibrium Throughout Matsoukas focuses on topics that link tightly to other key areas of undergraduate chemical engineering including separations reactions and capstone design More than 300 end of chapter problems range from basic calculations to realistic environmental applications these can be solved with any leading mathematical software Coverage includes Pure fluids PVT behavior and basic calculations of enthalpy and entropy Fundamental relationships and the calculation of properties from equations of state Thermodynamic analysis of chemical processes Phase diagrams of binary and simple ternary systems Thermodynamics of mixtures using equations of state Ideal and nonideal solutions Partial miscibility solubility of gases and solids osmotic processes Reaction equilibrium with applications to single and multiphase reactions

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Chemical Process Equipment James R. Couper, W. Roy Penney, 2012-09-19 First published Chemical process equipment Stanley M Walas 1988

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