

Gravimetric Analysis Example – try it!

- An excess of aqueous sodium phosphate is added to a 50.0 mL sample of copper (II) chloride, with unknown concentration. A precipitate forms, as shown in the molecular equation below:



A student purifies, dries, and weighs the solid precipitate. She weighs the solid and finds she has collected 18.0 grams of $\text{Cu}_3(\text{PO}_4)_2$. Based on this data, determine the molarity of the CuCl_2 solution.

$$18.0 \text{ g } \cancel{\text{Cu}_3(\text{PO}_4)_2} \times \frac{1 \text{ mol } \cancel{\text{Cu}_3(\text{PO}_4)_2}}{380.4 \text{ g } \cancel{\text{Cu}_3(\text{PO}_4)_2}} \times \frac{3 \text{ mol } \text{CuCl}_2}{1 \cancel{\text{Cu}_3(\text{PO}_4)_2}} = 0.142 \text{ mol } \text{CuCl}_2$$

$$M = \frac{\text{mol}}{\text{L}} = \frac{0.142 \text{ mol}}{0.0500 \text{ L}} = 2.84$$

Gravimetric Analysis Lab Calculations

Ying-Ying Zheng



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English in Analytical Chemistry. Communicating about Methods & Techniques. Книга для студента Надежда Зорина, Александра Соболева, 2022-10-19 Catalogue Yale University, 1915 *Laboratory Experiments for Brown and LeMay, Chemistry, the Central Science* John Henry Nelson, Kenneth C. Kemp, 1977 **University Catalogue** Yale University, 1909 *Catalogue of the Officers and Students in Yale College* Yale University, 1917 *Basic Laboratory Principles in General Chemistry* Fitzgerald B. Bramwell, 1990 *Report summaries* United States. Environmental Protection Agency, 1983 *Host Bibliographic Record for Boundwith Item Barcode 30112067897766 and Others*, 1906 **Analytical Chemistry for Technicians** John Kenkel, 2002-10-29 Surpassing its bestselling predecessors this thoroughly updated third edition is designed to be a powerful training tool for entry level chemistry technicians Analytical Chemistry for Technicians Third Edition explains analytical chemistry and instrumental analysis principles and how to apply them in the real world A unique feature of this edition is that it brings the workplace of the chemical technician into the classroom With over 50 workplace scene sidebars it offers stories and photographs of technicians and chemists working with the equipment or performing the techniques discussed in the text It includes a supplemental CD that enhances training activities The author incorporates knowledge gained from a number of American Chemical Society and PITTCON short courses and from personal visits to several laboratories at major chemical plants where he determined firsthand what is important in the modern analytical laboratory The book includes more than sixty experiments specifically relevant to the laboratory technician along with a Questions and Problems section in each chapter Analytical Chemistry for Technicians Third Edition continues to offer the nuts and bolts of analytical chemistry while focusing on the practical aspects of training **Catalogue of the Officers and Graduates of Yale University** Yale University, 1909 School of Mines University of Minnesota. School of Mines, 1911 *Catalogue of the University of Dakota for the Year ..., with Announcements for the Year ...* University of South Dakota, 1919 **Biennial Report of the General Extension Division, University of Minnesota** University of Minnesota. General Extension Division, 1923 **Catalog** University of South Dakota, 1917 **Annual Register of the State University of Nevada ... with Announcements ...** University of Nevada, 1905 *Catalogue ...* Yale University. Sheffield Scientific School, 1905 **Colorado State College Bulletin** Colorado State College (Fort Collins, Colo.), 1915 **Bulletin** New York University, 1918 Catalogue Number New York State College of Ceramics at Alfred University, 1922 **The Temple University Catalogue** Temple University, 1913

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