



FOURTH EDITION

GEOGRAPHIC INFORMATION SCIENCE AND SYSTEMS

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web qualitative test flame tests introduction for this test i had five containers labelled a e each containing a metal salt i had to find out which metal is used in each of the five unknown salts by using a metal loop placing it in the unknown salts and testing it by putting the metal under a roaring bunsen flame safety points wear goggles all the time when

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web the easiest colors to observe during the flame test were that of copper chloride and lithium chloride for potassium chloride there was a purple color close to the stick but the flame remained orange for the majority of the burn time the unknown slat burned orange but also had a slight blue color close to the stick

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web use crayons or colored pencils to color in the spectrum in the box on your report sheet to show the correlation between

color and wavelength part ii flame tests you will conduct flame tests to observe the flame emission colors for the following solutions LiNO_3 CuNO_3 2SrNO_3 2BaNO_3 2KNO_3 and NaNO_3 since nitrates do not emit color you

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metal cations in this lab how do the metal cations become excited

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