

Chapter Test B

Chapter Chemical Equations and Reactions

Write the full name (including units) for the term at the end of the sentence. Please show your work. Please write answers without answers with symbols.

- The molar volume of an ideal gas at standard temperature and pressure is 22.7 L mol^{-1} .
 - gas
 - pressure
 - standard temperature
 - volume
- The volume of a gas is independent of its pressure at constant temperature.
 - constant
 - volume
 - standard temperature
 - number of particles
- An unknown substance has a molar mass of 100 g mol^{-1} .
 - mass
 - density of water
 - volume
 - number
- The molar mass of a compound is 100 g mol^{-1} .
 - standard temperature and pressure
 - density of water
 - density of the compound
 - standard temperature and pressure at standard volume
- The molar mass of a compound is 100 g mol^{-1} .
 - density of water
 - density of the compound
 - density of the compound
 - density of the compound
- The molar mass of a compound is 100 g mol^{-1} .
 - density of water
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 - density of the compound

Chapter Test B Chemical Equations And Reactions Answers

Christian G. Meyer



Chapter Test B Chemical Equations And Reactions Answers:

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