

19

ACIDS, BASES, AND SALTS

SECTION 19.1 ACID-BASE THEORIES (pages 587–593)

This section compares and contrasts acids and bases as defined by the theories of Arrhenius, Brønsted-Lowry, and Lewis. It also identifies conjugate acid-base pairs in acid-base reactions.

► Properties of Acids and Bases (pages 587–588)

- Circle the letters of all the terms that complete the sentence correctly.
The properties of acids include _____ .
 - reacting with metals to produce oxygen
 - giving foods a sour taste
 - forming solutions that conduct electricity
 - causing indicators to change color
- Bases are compounds that react with acids to form _____
and a(n) _____ .
- Circle the letters of all the terms that complete the sentence correctly.
The properties of bases include _____ .
 - tasting bitter
 - feeling slippery
 - changing the color of an indicator
 - always acting as a strong electrolyte

► Arrhenius Acids and Bases (pages 588–590)

- Match the number of ionizable hydrogens with the type of acid.

_____ one	a. diprotic
_____ two	b. triprotic
_____ three	c. monoprotic
- Is the following sentence true or false? Only the hydrogens in weak polar bonds are ionizable. _____
- Hydrogen is joined to a very _____ element in a very polar bond.
- Alkali metals react with water to produce _____ solutions.

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Table of Contents Chapter 20 Acids And Bases Answer Key

1. Understanding the eBook Chapter 20 Acids And Bases Answer Key
 - The Rise of Digital Reading Chapter 20 Acids And Bases Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 20 Acids And Bases Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 20 Acids And Bases Answer Key
 - User-Friendly Interface

4. Exploring eBook Recommendations from Chapter 20 Acids And Bases Answer Key
 - Personalized Recommendations
 - Chapter 20 Acids And Bases Answer Key User Reviews and Ratings
 - Chapter 20 Acids And Bases Answer Key and Bestseller Lists
5. Accessing Chapter 20 Acids And Bases Answer Key Free and Paid eBooks
 - Chapter 20 Acids And Bases Answer Key Public Domain eBooks
 - Chapter 20 Acids And Bases Answer Key eBook Subscription Services
 - Chapter 20 Acids And Bases Answer Key Budget-Friendly Options
6. Navigating Chapter 20 Acids And Bases Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 20 Acids And Bases Answer Key Compatibility with Devices
 - Chapter 20 Acids And Bases Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 20 Acids And Bases Answer Key
 - Highlighting and Note-Taking Chapter 20 Acids And Bases Answer Key
 - Interactive Elements Chapter 20 Acids And Bases Answer Key
8. Staying Engaged with Chapter 20 Acids And Bases Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 20 Acids And Bases Answer Key
9. Balancing eBooks and Physical Books Chapter 20 Acids And Bases Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 20 Acids And Bases Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 20 Acids And Bases Answer Key
 - Setting Reading Goals Chapter 20 Acids And Bases Answer Key
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Chapter 20 Acids And Bases Answer Key
 - Fact-Checking eBook Content of Chapter 20 Acids And Bases Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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