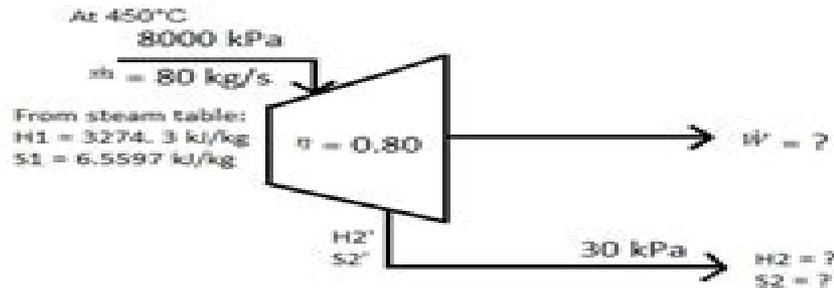


Introduction to Chemical Engineering Thermodynamics
Problem Set

1. A turbine operates adiabatically with superheated steam entering at 450°C and 8,000 kPa with a mass flowrate of 80 kg/s. The exhaust pressure is 30 kPa and the turbine efficiency is 0.80. Determine the power output of the turbine and the enthalpy and entropy of the exhaust steam.

Given:



Required:

- \dot{W} , power output of the turbine
- H_2 , enthalpy of exhaust stream
- S_2 , entropy of exhaust stream

Solution:

From steam table under superheated steam, Table F.2, for inlet conditions:

$$H_1 = 3274.3 \frac{\text{kJ}}{\text{kg}}$$

$$S_1 = 6.5597 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}$$

For discharge conditions:

$$H_2^L = 289.302 \frac{\text{kJ}}{\text{kg}}, H_2^V = 2625.4 \frac{\text{kJ}}{\text{kg}}$$

$$S_2^L = 0.9441 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}, S_2^V = 7.7695 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}$$

If the expansion at 30 kPa is isentropic, then: $S_2^L = S_1 = 6.5597 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}$. Steam with this entropy at 10 kPa is wet, yielding the equation:

$$S_2^L = S_2^L + x_2'(S_2^V - S_2^L)$$

Substituting the values:

$$6.5597 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} = 0.9441 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} + x_2' \left(7.7695 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} - 0.9441 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} \right)$$

$$x_2' = 0.8228$$

This is the quality (fraction vapor) of the discharge stream at point 2'. The enthalpy H_2^L is also given by the equation:

$$H_2^L = H_2^L + x_2'(H_2^V - H_2^L)$$

Thus,

Chemical Engineering Thermodynamics Problems And Solutions

RC Schank



Chemical Engineering Thermodynamics Problems And Solutions:

This book delves into Chemical Engineering Thermodynamics Problems And Solutions. Chemical Engineering Thermodynamics Problems And Solutions is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chemical Engineering Thermodynamics Problems And Solutions, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Chemical Engineering Thermodynamics Problems And Solutions
 - Chapter 2: Essential Elements of Chemical Engineering Thermodynamics Problems And Solutions
 - Chapter 3: Chemical Engineering Thermodynamics Problems And Solutions in Everyday Life
 - Chapter 4: Chemical Engineering Thermodynamics Problems And Solutions in Specific Contexts
 - Chapter 5: Conclusion
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 3. In chapter 2, this book will delve into the foundational concepts of Chemical Engineering Thermodynamics Problems And Solutions. The second chapter will elucidate the essential principles that need to be understood to grasp Chemical Engineering Thermodynamics Problems And Solutions in its entirety.
 4. In chapter 3, this book will examine the practical applications of Chemical Engineering Thermodynamics Problems And Solutions in daily life. The third chapter will showcase real-world examples of how Chemical Engineering Thermodynamics Problems And Solutions can be effectively utilized in everyday scenarios.
 5. In chapter 4, this book will scrutinize the relevance of Chemical Engineering Thermodynamics Problems And Solutions in specific contexts. The fourth chapter will explore how Chemical Engineering Thermodynamics Problems And Solutions is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, this book will draw a conclusion about Chemical Engineering Thermodynamics Problems And Solutions. The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Chemical Engineering Thermodynamics Problems And Solutions.

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