

Physical Chemistry

Revised And Enlarged Seventh Edition

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Revised by
S. C. Rakshit

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H. K. MOUDGIL



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Physical Chemistry H. L. Heys, 1987 **Textbook of Physical Chemistry** H. K. MOUDGIL, 2010-07 This comprehensive textbook covers the principal areas of physical chemistry such as thermodynamics quantum chemistry molecular spectroscopy chemical kinetics electrochemistry and nanotechnology In a methodical and accessible style the book discusses classical irreversible and statistical thermodynamics and statistical mechanics and describes macroscopic chemical systems steady states and thermodynamics at a molecular level It elaborates the underlying principles of quantum mechanics

molecular spectroscopy X ray crystallography and solid state chemistry along with their applications The book explains various instrumentation techniques such as potentiometry polarography voltametry conductometry and coulometry It also describes kinetics rate laws and chemical processes at the electrodes In addition the text deals with chemistry of corrosion and nanomaterials This book is primarily designed for the undergraduate and postgraduate students of chemistry B Sc and M Sc for courses in physical chemistry Key Features Gives a thorough treatment to ensure a solid grasp of the material Presents a large number of figures and diagrams that help amplify key concepts Contains several worked out examples for better understanding of the subject matter Provides numerous chapter end exercises to foster conceptual understanding

A Textbook of Physical Chemistry - Volume 1 Mandeep Dalal, 2018-01-01 An advanced level textbook of physical chemistry for the graduate B Sc and postgraduate M Sc students of Indian and foreign universities This book is a part of four volume series entitled A Textbook of Physical Chemistry Volume I II III IV CONTENTS Chapter 1 Quantum Mechanics I Postulates of quantum mechanics Derivation of Schrodinger wave equation Max Born interpretation of wave functions The Heisenberg s uncertainty principle Quantum mechanical operators and their commutation relations Hermitian operators elementary ideas quantum mechanical operator for linear momentum angular momentum and energy as Hermitian operator The average value of the square of Hermitian operators Commuting operators and uncertainty principle x E Schrodinger wave equation for a particle in one dimensional box Evaluation of average position average momentum and determination of uncertainty in position and momentum and hence Heisenberg s uncertainty principle Pictorial representation of the wave equation of a particle in one dimensional box and its influence on the kinetic energy of the particle in each successive quantum level Lowest energy of the particle Chapter 2 Thermodynamics I Brief resume of first and second Law of thermodynamics Entropy changes in reversible and irreversible processes Variation of entropy with temperature pressure and volume Entropy concept as a measure of unavailable energy and criteria for the spontaneity of reaction Free energy enthalpy functions and their significance criteria for spontaneity of a process Partial molar quantities free energy volume heat concept Gibb s Duhem equation Chapter 3 Chemical Dynamics I Effect of temperature on reaction rates Rate law for opposing reactions of 1st order and 2nd order Rate law for consecutive Collision theory of reaction rates and its limitations Steric factor Activated complex theory Ionic reactions single and double sphere models Influence of solvent and ionic strength The comparison of collision and activated complex theory Chapter 4 Electrochemistry I Ion Ion Interactions The Debye Huckel theory of ion ion interactions Potential and excess charge density as a function of distance from the central ion Debye Huckel reciprocal length Ionic cloud and its contribution to the total potential Debye Huckel limiting law of activity coefficients and its limitations Ion size effect on potential Ion size parameter and the theoretical mean activity coefficient in the case of ionic clouds with finite sized ions Debye Huckel Onsager treatment for aqueous solutions and its limitations Debye Huckel Onsager theory for non aqueous solutions The solvent effect on the mobility at infinite dilution Equivalent

conductivity vs concentration c_1^2 as a function of the solvent Effect of ion association upon conductivity Debye Huckel Bjerrum equation Chapter 5 Quantum Mechanics II Schrodinger wave equation for a particle in a three dimensional box The concept of degeneracy among energy levels for a particle in three dimensional box Schrodinger wave equation for a linear harmonic oscillator Zero point energy of a particle possessing harmonic motion and its consequence Schrodinger wave equation for three dimensional Rigid rotator Energy of rigid rotator Space quantization Schrodinger wave equation for hydrogen atom separation of variable in polar spherical coordinates and its solution Principle azimuthal and magnetic quantum numbers and the magnitude of their values Probability distribution function Radial distribution function Shape of atomic orbitals s p Law of mass action and its thermodynamic derivation Third law of thermodynamics Nernst heat theorem determination of absolute entropy unattainability of absolute zero and its limitation Phase diagram for two completely miscible components systems Eutectic systems Calculation of eutectic point Systems forming solid compounds Ax By with congruent and incongruent melting points Phase diagram and thermodynamic treatment of solid solutions Chapter 7 Chemical Dynamics II Chain reactions hydrogen bromine reaction pyrolysis of acetaldehyde decomposition of ethane Photochemical reactions hydrogen bromine General treatment of chain reactions ortho para hydrogen conversion and hydrogen bromine reactions Apparent activation energy of chain reactions Chain length Rice Herzfeld mechanism of organic molecules decomposition acetaldehyde Branching chain reactions and explosions H₂ O₂ reaction Kinetics of one intermediate enzymatic reaction Michaelis Menton treatment Evaluation of Michaelis s constant for enzyme substrate binding by Lineweaver Burk plot and Eadie Hofstae methods Competitive and non competitive inhibition Chapter 8 Electrochemistry II Ion Transport in Solutions Ionic movement under the influence of an electric field Mobility of ions Ionic drift velocity and its relation with current density Einstein relation between the absolute mobility and diffusion coefficient The Stokes Einstein relation The Nernst Einstein equation Walden s rule The Rate process approach to ionic migration The Rate process equation for equivalent conductivity Total driving force for ionic transport Nernst Planck Flux equation Ionic drift and diffusion potential the Onsager phenomenological equations The basic equation for the diffusion Planck Henderson equation for the diffusion potential Advanced Practical Physical Chemistry Yasmin Akhtar, Mohammad Yasin, 2017-03-19

Advanced Physical Chemistry Mehra Harish C, 1978 *A Textbook of Physical Chemistry (Vol. 5)* K L Kapoor, 2004-02 Volume five covers dynamics of Chemical Reactions Statistical Thermodynamics and Macromolecules in five chapters such as Adsorption Chemical Kinetics Photo chemistry Statistical Thermodynamics and Macromolecules Physical Chemistry: a Molecular Approach McQuarrie Donald A, Simon John D, 1988 *Physical Chemistry in Brief* Ing. Anatol Malijevisky, 2014-09-19 The Physical Chemistry In Brief offers a digest of all major formulas terms and definitions needed for an understanding of the subject They are illustrated by schematic figures simple worked out examples and a short accompanying text The concept of the book makes it different from common university or physical chemistry textbooks *Textbook of*

Physical Chemistry N. K. Vishnoi, 2010 *Textbook of Physical Chemistry* L.K. Sharma, K.K. Sharma, 1990 Physical Chemistry Donald A. McQuarrie, John D. Simon, 2013 *Physical Chemistry* M. V. Sangaranarayanan, V. Mahadevan, 2012-04-16

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