

CHIMICA

STRUTTURA, COMPOSIZIONE, TRASFORMAZIONI:
DELLA MATERIA

↓
SI STUDIANO ATTRAVERSO IL
MODELLO ATOMICO

TUTTI GLI ATOMI DI
UNA DATA SPECIE SONO UGUALI MA
DIFFERISCONO PER LE LORO
CARATTERISTICHE QUANTITATIVE
(dal n° di protoni, elettroni e neutroni)

↳ ATOMO (α-τέμνω)

È IL PIÙ PICCOLO SISTEMA
ORGANIZZATO LA CUI NATURA,
TIPO DI LEGAME E LA SUA
DISTRIBUZIONE DETERMINANO
LE PROPRIETÀ MACROSCOPICHE
DELLA MATERIA

ESISTONO DUE TIPOLOGIE
DI SPECIE CHIMICHE
(O SOSTANZE CHIMICHE PURE)

SOSTANZE
ELEMENTARI

↓
SPECIE CHIMICHE FORMATE
DA ATOMI TUTTI UGUALI

INSIEMI O SISTEMI OMOGENEI ED
ETEROGENEI FORMATI DA SPECIE
CHIMICHE (RAPPORTI TRA ATOMI)

COMPOSTI

↓
SPECIE CHIMICHE FORMATE
DA DUE O PIÙ ATOMI DIVERSI
IN RAPPORTI QUANTITATIVI
CONSTANTI

C'È UNO DEI CRITERI DI DISCRIMINAZIONE
TRA QUESTE DUE SPECIE CHIMICHE CHE
RIGUARDANO IL TIPO DI INTERAZIONE TRA GLI ATOMI

- ATOMI SINGOLI NON INTERAGENTI
- MOLECOLE
- INSIEMI CONTINUI DI ATOMI

- MOLECOLE (COMBINANO TUTTA LA NATURA BIOLOGICA) !
- INSIEMI CONTINUI DI ATOMI DIVERSI
- IONI DI CARICA OPPOSTA
↳ EX ATOMI CON UN ELETTRONE IN PIÙ O IN MENO (LA NATURA FISICA)

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Petrucci



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