

Introduction 11

An Electron's Address

Name: _____
 Date: _____
 Class: _____

Introduction: Energy Levels and Electrons

The electrons in the following model would have populated the following lowest energy levels. The electrons would fill the lowest one's first, then the next one.

Remember that lowest energy levels are the most fundamental lowest energy levels. Each successive level is higher than the following one.

Figure 1

| Energy Level | Number of electrons in the lowest energy level |
|--------------|--|
| 1 | 2 |
| 2 | 8 |
| 3 | 18 |
| 4 | 32 |

How many electrons would be in a "fundamental" model of the 1st energy level because there would only be 2 electrons in the 1st energy level?

Critical Thinking Questions

- How many electrons would be in the 1st energy level?
 The fundamental model of the 1st energy level is 2 electrons in the lowest energy level.
- How many electrons would be in the 2nd energy level?
 The fundamental model of the 2nd energy level is 8 electrons in the lowest energy level.
- How many electrons would be in the 3rd energy level?
 The fundamental model of the 3rd energy level is 18 electrons in the lowest energy level.
- How many electrons would you expect to have in the 4th energy level?
 The number of electrons I would expect to have in the 4th energy level would be 32 electrons. There would be 18 electrons in the 3rd energy level and 14 electrons in the 4th energy level.
- How many electrons would you expect to have in the 5th energy level?
 The number of electrons I would expect to have in the 5th energy level would be 50 electrons. There would be 32 electrons in the 4th energy level and 18 electrons in the 5th energy level.

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**American Chemical Society.
Meeting, American Chemical Society**

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Database ,1989 **Online** ,1989 *Comprehensive Medicinal Chemistry* Corwin Hansch,Peter George Sammes,John Bodenhan Taylor,1990 Drug design is a multi disciplinary activity involving chemists biologists bochemists mpharmacologists and many others the chemist s role is central in inventing new compounds which exert a beneficial effect However once a lead for a new active drug has been established its effective delivery has to be demonstrated and extensive toxikological studies undertaken to demonstrate its safety before cinical trials can commence The metabolic fate of the drug has to be revealed and detailed distribution studies carried out in order to satisfy the regulatory authorities before the new compound can be marketed *Comprehensive Medicinal Chemistry* describes all these aspects of the design of a drug whilst centering on the chemical mechanism whereby such agents act Volume 4 covers quantitaive drug design **Chemical Week** ,2000

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