

## Chapter Test II

### A. Matching

Match each item in Column I with the correct description in Column II. Write the letter of the correct description.

#### Column I

1. An oxidation-reduction reaction in which an element is oxidized and another element is reduced.
2. An equation of a redox reaction that is balanced for both mass and charge.
3. The reduction of an oxidizing agent.
4. The ratio of the amount of substance that is oxidized to the amount of substance that is reduced.
5. An oxidation-reduction reaction in which the oxidizing agent is also the reducing agent.
6. An oxidation-reduction reaction in which the oxidizing agent is reduced and the reducing agent is oxidized.

#### Column II

- a. oxidation
- b. limiting reagent
- c. balanced redox
- d. stoichiometry
- e. redox reaction
- f. oxidizing agent

### B. Multiple Choice

Choose the best answer for each question.

1. Which of the following is not a redox reaction?
  - a.  $\text{Zn} + \text{Cu}^{2+} \rightarrow \text{Zn}^{2+} + \text{Cu}$
  - b.  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
  - c.  $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
  - d.  $\text{H}_2 + \text{CO}_2 \rightarrow \text{H}_2\text{O} + \text{C}$
2. Which of the following is not a redox reaction?
  - a.  $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
  - b.  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
  - c.  $\text{H}_2 + \text{CO}_2 \rightarrow \text{H}_2\text{O} + \text{C}$
  - d.  $\text{Zn} + \text{Cu}^{2+} \rightarrow \text{Zn}^{2+} + \text{Cu}$
3. Which of the following is not a redox reaction?
  - a.  $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
  - b.  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
  - c.  $\text{H}_2 + \text{CO}_2 \rightarrow \text{H}_2\text{O} + \text{C}$
  - d.  $\text{Zn} + \text{Cu}^{2+} \rightarrow \text{Zn}^{2+} + \text{Cu}$

# Chemistry Chapter 12 Stoichiometry Test

**Lei Shi**



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