

Important Questions for Class 9
Science

Chapter 3 – Atoms and Molecules

Very Short Answer Questions

1 Mark

1. Atomic radius is measured in nanometers and

- (a) $1\text{nm} = 10^{-10}\text{m}$
- (b) $1\text{m} = 10^{-10}\text{nm}$
- (c) $1\text{nm} = 10^{-9}\text{m}$
- (d) $1\text{m} = 10^{-9}\text{nm}$

Ans: (c) $1\text{nm} = 10^{-9}\text{m}$

2. Symbol of Iron is –

- (a) Ir
- (b) I
- (c) Fe
- (d) None of these

Ans: (c) Fe

3. Atomicity of Chlorine and Argon is

- (a) Diatomic and Monoatomic
- (b) Monoatomic and Diatomic
- (c) Monoatomic and Monoatomic
- (d) Diatomic and Diatomic

Ans: (a) Diatomic and Monoatomic

4. Molecular mass of water (H_2O) is

- (a) 18g
- (b) 8g
- (c) 33g
- (d) 34g

Ans: (a) 18g

5. It is said that 1 mole of a compound contains –

- (a) 6.023×10^{23} atoms
- (b) 6.023×10^{24} atoms
- (c) 60.23×10^{23} atoms

Chapter 3 Atoms And Moles

**Mary Summerfield Gardiner, Sarah C.
Flemister**

Chapter 3 Atoms And Moles:

The Enigmatic Realm of **Chapter 3 Atoms And Moles**: Unleashing the Language is Inner Magic

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