

Assessment

Chapter Test B**Teacher Notes and Answers****8** *Chemical Equations and Reactions***TEST B**

- b
- a
- b
- d
- a
- b
- d
- a
- coefficient
- liquid
- reversible reaction
- $A + X \rightarrow AX$
- $AX \rightarrow A + X$
- $A + BX \rightarrow AX + B$
- $AX + BY \rightarrow AY + BX$
- metal hydroxides
- combustion reaction
- electrolysis
- activity series
- chemical equation
- conservation of mass
- gaseous
- heated
- A word equation and a formula equation are both qualitative. Although they both show the reactants and products of a chemical reaction, they give no information about amounts of reactants or products. A chemical equation, however, does show the relative amounts of the reactants and products in a chemical reaction.
- Three indications of a chemical reaction are the radiation of heat and light, the production of a gas, and the formation of a precipitate.
- zinc sulfide + oxygen \rightarrow
zinc oxide + sulfur dioxide
- sodium iodide + chlorine \rightarrow
sodium chloride + iodine
- $Mg + O_2 \rightarrow MgO$
- $CaCO_3 \rightarrow CaO + CO_2$
- decomposition reaction
- single-displacement reaction
- synthesis reaction
- double-displacement reaction
- $2AgNO_3 + CuCl_2 \rightarrow 2AgCl + Cu(NO_3)_2$
- $2PbO_2 \rightarrow 2PbO + O_2$
- $Zn(OH)_2 + 2CH_3COOH \rightarrow$
 $Zn(CH_3COO)_2 + 2H_2O$
- $Mg(s) + 2H_2O(g) \rightarrow Mg(OH)_2(aq) + H_2(g)$
- no reaction
- $Cl_2(g) + MgBr_2(aq) \rightarrow MgCl_2(aq) + Br_2(l)$
- $Zn(s) + 2HCl(aq) \rightarrow ZnCl_2(aq) + H_2(g)$
- no reaction
- $H_2(s) + CuCl_2(aq) \rightarrow H_2Cl_2(aq) + Cu(s)$
- no reaction
- $Mg(s) + Co(NO_3)_2(aq) \rightarrow$
 $Mg(NO_3)_2(aq) + Co(s)$

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Victor M. Corman



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Kaplan SAT Subject Test Chemistry 2015-2016 Kaplan Test Prep, 2015-03-03 Essential strategies practice and review to ace the SAT Subject Test Chemistry Getting into a top college has never been more difficult Students need to distinguish themselves from the crowd and scoring well on a SAT Subject Test gives students a competitive edge Kaplan's SAT Subject Test Chemistry is the most up to date guide on the market with complete coverage of both the content review and strategies students need for success on test day Kaplan's SAT Subject Test Chemistry features A full length diagnostic test Full length practice tests Focused chapter summaries highlights and quizzes Detailed answer explanations Proven score raising strategies End of chapter quizzes Kaplan is serious about raising students scores we guarantee students will get a higher score

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Demosthenes,1898 *MCAT 45, 2004* Kaplan, Inc,2004-02 This new edition offers hundreds of the toughest practice questions to help candidates prepare for the MCAT

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