

To find the probability of Two Dependent Events:

$$P(A \text{ and } B) = P(A) \cdot P(B \text{ after } A)$$

probability  
of both events

probability  
of first event

probability of second event  
after the first event has occurred

A bag contains 2 black marbles and 3 white marbles.  
Find the probability of drawing a black marble and putting  
it aside and then a drawing a white marble.



$$P(\text{black and white}) = P(\text{black}) \cdot P(\text{white after black})$$

$$P(\text{black and white}) = \frac{2}{5} \cdot \frac{3}{4}$$

$$P(\text{black and white}) = \frac{6}{20} = \frac{3}{10}$$

The probability of drawing a black marble and then a white marble is  $\frac{3}{10}$ .

Now you try: Find the probability of each event when after the first draw you do not return the marble to the bag.

1.  $P(\text{white and black})$  \_\_\_\_\_

$$\frac{3}{5} \cdot \frac{2}{4} = \frac{6}{20} = \frac{3}{10}$$

2.  $P(2 \text{ white marbles})$  \_\_\_\_\_

$$\frac{3}{5} \cdot \frac{2}{4} = \frac{6}{20} = \frac{3}{10}$$

*(You can cross reduce before multiplying!)*

Nicole has 10 coins in her purse: 3 pennies, 3 nickels, 2 dimes, and 2 quarters.  
Nicole draws one coin from her purse and puts it on the counter.

3. What is the probability that Nicole will draw a penny and a dime?

$$P(\text{penny and dime}) =$$

$$\frac{3}{10} \cdot \frac{2}{9} = \frac{6}{90} = \frac{1}{15}$$

4. What is the probability that Nicole will draw a quarter and a nickel?

$$P(\text{quarter and nickel}) =$$

$$\frac{2}{10} \cdot \frac{3}{9} = \frac{6}{90} = \frac{1}{15}$$

# Independent And Dependent Probability Worksheet With Answer Key

**David S. Moore, George P.  
McCabe, Layth C. Alwan, Bruce A.  
Craig, William M. Duckworth**

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**The Practice of Statistics for Business and Economics** David S. Moore,George P. McCabe,Layth C. Alwan,Bruce A. Craig,William M. Duckworth,2010-09-15 This innovative textbook is designed to give students the tools they need to make data informed real world business decisions practically from the first day of class providing a foundation in data production

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the powdered acetamide occupies 1 2 mm of the capillary tube end c

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the melting point apparatus compound a had a melting point range of 114.6 - 115.5 c thus revealing it to be acetanilide compound b had a melting point range of 122.6 - 124.5 c thus

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 °C so in order to convert an experimental measurement taken in Calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that needs to be determined by physical properties such as boiling point melting point density solubility etc in this experiment I aim to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a Bunsen burner boiling

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