

CHEMISTRY



Atomic Structure

Atoms are the smallest particles which cannot be created or destroyed in a chemical reaction. They are made up of three sub-particles: electrons, protons and neutrons.

Atomic Number

The number of protons present in the nucleus of an atom is called its atomic number. It is denoted by the symbol 'Z'.

Atomic Mass

The mass of an atom is called its atomic mass. It is denoted by the symbol 'A'.

Isotopes

Atoms of the same element having the same atomic number but different mass numbers are called isotopes.

Atomic Structure

Atoms are made up of three sub-particles: electrons, protons and neutrons. Electrons are negatively charged, protons are positively charged, and neutrons are neutral.

Atomic Number

The number of protons present in the nucleus of an atom is called its atomic number. It is denoted by the symbol 'Z'.

Atomic Mass

The mass of an atom is called its atomic mass. It is denoted by the symbol 'A'.

Atomic Structure

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Element	Atomic Number (Z)	Atomic Mass (A)
Hydrogen	1	1
Helium	2	4
Lithium	3	7
Boron	5	11
Carbon	6	12
Nitrogen	7	14
Oxygen	8	16
Fluorine	9	19
Neon	10	20

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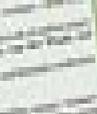
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Mark Jackson



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Sociology BarCharts, Inc,2014

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