

Naming Polyatomic Ionic Compounds Worksheet

Name the following ions.

1) $(\text{NH}_4)_2\text{CrO}_4$ _____

2) AgNO_3 _____

3) $\text{Al}(\text{ClO}_3)_3$ _____

4) $\text{Ba}(\text{ClO}_4)_2$ _____

5) CaSO_4 _____

6) $\text{Cd}(\text{OH})_2$ _____

7) $\text{Cu}(\text{ClO}_4)_2$ _____

8) $\text{Fe}(\text{HSO}_3)_2$ _____

9) $\text{Fe}(\text{NO}_2)_2$ _____

10) $\text{Hg}_2(\text{NO}_3)_2$ _____

11) HgCrO_4 _____

12) $\text{K}_2\text{Cr}_2\text{O}_7$ _____

13) $\text{Mg}_3(\text{PO}_4)_2$ _____

14) $\text{Pb}(\text{CO}_3)_2$ _____

15) $\text{Pb}(\text{SO}_4)_2$ _____

16) $\text{Pb}_3(\text{PO}_4)_2$ _____

17) $\text{Sn}(\text{CN})_2$ _____

18) $\text{Sn}(\text{CO}_3)_2$ _____

19) $\text{Sr}(\text{NO}_3)_2$ _____

20) $\text{Zn}(\text{HCO}_3)_2$ _____

Chemistry Ionic Compounds Polyatomic Ions Worksheet Answers

SB Merriam



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