

Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 3

Q1. In a hydrocarbon, the mass ratio of hydrogen and carbon is 1:3; the empirical formula of hydrocarbon is _____.

- a.) CH_4
- b.) CH_2
- c.) C_2H
- d.) CH_3

Q2. 80g of oxygen contains as many atoms as in _____.

- a.) 10 g of hydrogen
- b.) 5 g of hydrogen
- c.) 8 g of hydrogen
- d.) 1 g of hydrogen

Q3. A mixture of gases contains H_2 and O_2 gases in the ratio 1:4. What is the molar ratio of the two gases in the mixture?

- a.) 16:1
- b.) 2:1
- c.) 1:4
- d.) 4:1

Q4. Which of the following concentration terms is/are independent of temperature?

- a.) Molality
- b.) Mole fraction
- c.) Molarity
- d.) Normality

Q5. Which of the following contains the highest number of atoms?

- a.) 1.0 g of water
- b.) 1.0 g of silver
- c.) 1.0 g of nitrogen
- d.) 1.0 g of propane

Chemistry Chapter 17 Assessment Answers

Ian Pickup



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Table of Contents Chemistry Chapter 17 Assessment Answers

1. Understanding the eBook Chemistry Chapter 17 Assessment Answers
 - The Rise of Digital Reading Chemistry Chapter 17 Assessment Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 17 Assessment Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Chapter 17 Assessment Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Chapter 17 Assessment Answers
 - Personalized Recommendations
 - Chemistry Chapter 17 Assessment Answers User Reviews and Ratings
 - Chemistry Chapter 17 Assessment Answers and Bestseller Lists
5. Accessing Chemistry Chapter 17 Assessment Answers Free and Paid eBooks

- Chemistry Chapter 17 Assessment Answers Public Domain eBooks
 - Chemistry Chapter 17 Assessment Answers eBook Subscription Services
 - Chemistry Chapter 17 Assessment Answers Budget-Friendly Options
6. Navigating Chemistry Chapter 17 Assessment Answers eBook Formats
- ePub, PDF, MOBI, and More
 - Chemistry Chapter 17 Assessment Answers Compatibility with Devices
 - Chemistry Chapter 17 Assessment Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Chemistry Chapter 17 Assessment Answers
 - Highlighting and Note-Taking Chemistry Chapter 17 Assessment Answers
 - Interactive Elements Chemistry Chapter 17 Assessment Answers
8. Staying Engaged with Chemistry Chapter 17 Assessment Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Chapter 17 Assessment Answers
9. Balancing eBooks and Physical Books Chemistry Chapter 17 Assessment Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Chapter 17 Assessment Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Chapter 17 Assessment Answers
- Setting Reading Goals Chemistry Chapter 17 Assessment Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Chapter 17 Assessment Answers
- Fact-Checking eBook Content of Chemistry Chapter 17 Assessment Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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