

CHEMICAL REACTIONS

webquest

emergency sub plans!

CHEMICAL REACTIONS Name: _____

Learn about chemical reactions by reading the text below.

1. What happens to matter in a chemical reaction?

2. Write the equation for a chemical reaction? Explain.

3. What happens to the particles and products in a rxn?

4. Draw the diagram of the collision and showing if it will be successful or not.

5. What do you notice about the number of hydrogen atoms on each side of the reactants?

6. Complete the chart below with the four signs of rxn.

TYPE OF EVIDENCE	REACTIONS

CHEMICAL REACTIONS Name: _____

Learn about chemical reactions by reading the text below.

1. What is interesting about the composition of water?

2. Write a chemical equation representing the rxn that H₂ and O₂ are reacting to form?

3. What happens to the gases that react in this rxn?

4. How does water in the air get there? (Chemical rxn or not?)

5. What is the relationship between reactants and products?

6. Write a chemical equation for the rxn that produces water from H₂ and O₂.

7. What happens to the volume and color of the rxn?

8. Why does this rxn not take place in a closed container?

9. What are the signs of a chemical rxn? (Change in color?)

10. Write a chemical equation for the rxn that produces H₂O from H₂ and O₂.

11. Explain what happens in a combustion rxn.

CHEMICAL REACTIONS Name: _____

Practice balancing chemical reactions by filling in the table below.

1. $2H_2 + O_2 \rightarrow 2H_2O$

Reactants	Products
H	H
O	O

2. $2H_2 + O_2 \rightarrow 2H_2O$

Reactants	Products
H	H
O	O

3. $2H_2 + O_2 \rightarrow 2H_2O$

Reactants	Products
H	H
O	O

4. $2H_2 + O_2 \rightarrow 2H_2O$

Reactants	Products
H	H
O	O

CHEMICAL REACTIONS Name: _____

Learn about the signs of chemical reactions by reading the text below.

1. What does the use of concentration of reactants tell you?

2. Chemical reactions are exothermic _____ rxns.

3. In a chemical equation (CE), how many oxygen atoms are there? How do you get to this number?

4. In a chemical equation (CE), how many oxygen atoms are there? How do you get to this number?

5. Complete the chart with the number of oxygen atoms in each chemical equation. Then the relationship in your own words. Remember the idea of conservation of mass.

Chemical Equation	Reactants	Products

6. Complete the chart below with the number of oxygen atoms in each chemical equation. Then the relationship in your own words. Remember the idea of conservation of mass.

Chemical Equation	Reactants	Products



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Chemical Reactions Webquest Answers

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