



Chapter 3: Atoms And Molecules

Science Worksheet

Standard: 9th

Subject: Science

Student Name: _____

Q1. In a chemical reaction 10.6g of sodium carbonate reacted with 12g of ethanoic acid. The products were 4.4g carbon dioxide, 1.8g of water and sodium ethanoate. The mass of sodium ethanoate formed is

1. 16.4g
2. 0.16g
3. 24g
4. 8.2g

Q2. When two molecules of hydrogen react with one molecule of oxygen, the mass of reactants is 36, what is the mass of products:

1. 18
2. 36
3. 9
4. 32

Q3. Barium chloride reacts with sodium sulphate to form Barium sulphate and sodium chloride. Then, according to the law of conservation of mass:

1. The total mass of reactants, barium chloride, and sodium sulphate taken is greater than the total mass of products, barium sulphate, sodium chloride formed
2. The total mass of reactants, barium chloride, and sodium sulphate is less than the total mass of products, barium sulphate, and sodium chloride
3. The total mass of the reactants is equal to the total mass of products
4. None of the above

Q4. 'All matter is composed of very small particles called Parmanu' was first suggested by:

1. John Dalton
2. J.J. Thomson
3. Kanad
4. William J. Crooke

Q5. A particle X has 17 protons, 18 neutrons and 18 electrons. This particle is most likely to be:

1. A cation.



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