

Worksheet 2: Binary Ionic Compounds (variable charge metals) - metals from groups 3 - 12 and 14 or 4A (Pb and Sn) - charge of anion is constant - use charge of anion to figure out charge of metal

1. Name the following ionic compounds:

a. SnO_2	oxide = O^{-2}	Tin (IV) oxide
b. Mn_2O_7	oxide = O^{-2}	Manganese (VII) oxide
c. FeN	nitride = N^{-3}	Iron (III) nitride
d. Cu_3N_2	nitride = N^{-3}	Copper (II) nitride
e. TiF_3	fluoride = F^{-1}	Titanium (III) fluoride
f. Cu_2S	sulfide = S^{-2}	Copper (I) sulfide
g. Fe_2S_3	sulfide = S^{-2}	Iron (III) sulfide
h. CuBr	bromide = Br^{-1}	Copper (I) bromide
i. Co_3N_2	nitride = N^{-3}	Cobalt (II) nitride
j. CoF_2	fluoride = F^{-1}	Cobalt (II) fluoride

2. Write the formula for the following ionic compounds:

a. Copper (I) nitride	Cu_3N
b. Cobalt (I) fluoride	CoF
c. Titanium (IV) oxide	TiO_2
d. Iron (II) nitride	Fe_3N_2
e. Iron (III) chloride	FeCl_3
f. Copper (II) oxide	CuO
g. Rhodium (II) oxide	RhO
h. Tin (IV) sulfide	SnS_2
i. Manganese (IV) nitride	Mn_3N_4
j. Copper (I) fluoride	CuF
k. Cobalt (II) sulfide	CoS
l. Iron (III) oxide	Fe_3O_2

Writing And Naming Binary Compounds Worksheet Answer Key

Puja Mehta



Writing And Naming Binary Compounds Worksheet Answer Key:

Writing And Naming Binary Compounds Worksheet Answer Key Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Writing And Naming Binary Compounds Worksheet Answer Key**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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