

## Chemistry Chapter 8 Exam

### Multiple Choice

1. Which of the following is not a Lewis acid?

- \_\_\_\_\_ a.  $\text{CH}_3\text{COOH}$
- \_\_\_\_\_ b. Lewis Acids
- \_\_\_\_\_ c. Carbonyl Compounds
- \_\_\_\_\_ d.  $\text{BF}_3$
- \_\_\_\_\_ e. Lewis Acids
- \_\_\_\_\_ f.  $\text{H}^+$
- \_\_\_\_\_ g. Lewis Acids
- \_\_\_\_\_ h.  $\text{H}_2\text{O}$
- \_\_\_\_\_ i. Lewis Acids

- \_\_\_\_\_ j. Lewis Acids
- \_\_\_\_\_ k. Lewis Acids
- \_\_\_\_\_ l. Lewis Acids
- \_\_\_\_\_ m. Lewis Acids
- \_\_\_\_\_ n. Lewis Acids
- \_\_\_\_\_ o. Lewis Acids
- \_\_\_\_\_ p. Lewis Acids
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- \_\_\_\_\_ w. Lewis Acids
- \_\_\_\_\_ x. Lewis Acids
- \_\_\_\_\_ y. Lewis Acids
- \_\_\_\_\_ z. Lewis Acids

### Multiple Choice

2. The strongest Lewis acid is

- \_\_\_\_\_ a.  $\text{BF}_3$
- \_\_\_\_\_ b.  $\text{AlCl}_3$
- \_\_\_\_\_ c.  $\text{FeCl}_3$
- \_\_\_\_\_ d.  $\text{SnCl}_4$

3. The Lewis acid  $\text{BF}_3$  reacts with  $\text{NH}_3$  to form a complex. The Lewis acid is  $\text{BF}_3$  and the Lewis base is  $\text{NH}_3$ .

- \_\_\_\_\_ a.  $\text{BF}_3$
- \_\_\_\_\_ b.  $\text{NH}_3$
- \_\_\_\_\_ c.  $\text{BF}_3$
- \_\_\_\_\_ d.  $\text{NH}_3$

4. The Lewis acid  $\text{BF}_3$  reacts with  $\text{NH}_3$  to form a complex. The Lewis acid is  $\text{BF}_3$  and the Lewis base is  $\text{NH}_3$ .

- \_\_\_\_\_ a.  $\text{BF}_3$
- \_\_\_\_\_ b.  $\text{NH}_3$
- \_\_\_\_\_ c.  $\text{BF}_3$
- \_\_\_\_\_ d.  $\text{NH}_3$

# Modern Chemistry Chapter 6 Test Answers

**Joacim Rocklöv**



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