

Stoichiometry Worksheet 2
Chemistry

Name _____

Date _____ Hour _____

1. In the reaction : $2\text{Al}_2\text{O}_{3(s)} \rightarrow 4\text{Al}_{(s)} + 3\text{O}_{2(g)}$, the mole ratios are:

Al_2O_3 to Al _____ Al_2O_3 to O_2 _____ Al to O_2 _____ O_2 to Al _____

2. In the reaction: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, the mole ratio of oxygen to water is: _____

3. In the reaction: $\text{Ca} + \text{Cl}_2 \rightarrow \text{CaCl}_2$, the mole ratio of chlorine to calcium is: _____

4. In the reaction: $\text{Zn} + \text{H}_2\text{SO}_4 \rightarrow \text{ZnSO}_4 + \text{H}_2$, the mole ratio of zinc to sulfuric acid is:

5. The equation for the Haber process is represented by:



To convert 9.0 moles of hydrogen to ammonia would require how many moles of nitrogen?

6. In the equation: $2\text{KClO}_{3(g)} \rightarrow 2\text{KCl}_{(s)} + 3\text{O}_{2(g)}$, how many moles of oxygen are produced when 3.00 moles of KClO_3 decompose completely?

7. For the reaction: $\text{C}_{(s)} + 2\text{H}_{2(g)} \rightarrow \text{CH}_{4(g)}$, how many moles of H_2 are required to produce 10.0 moles of CH_4 ?

Chapter 9 Stoichiometry Section 2 Worksheet

**Donald J. Wink, Sharon Fetzer-
Gislason, Sheila McNicholas**

Chapter 9 Stoichiometry Section 2 Worksheet:

Stoichiometry and Thermodynamics of Metallurgical Processes Y. K. Rao, 1985-10-31 Originally published in 1985 this textbook provides a thorough and comprehensive coverage of a wide range of topics in stoichiometry and thermodynamics with special emphasis on applications to metallurgical processes This book will be welcomed as a text for courses in elementary and advanced thermodynamics and stoichiometry MCAT Comprehensive Review 1999, 1998-11 Complete Review of ALL the science you need to know Full Length Practice Test with complete explanations for each question 2 Full Length Verbal Reasoning Tests with detailed explanations and strategies *Handbook on Material and Energy Balance Calculations in Material Processing, Includes CD-ROM* Arthur E. Morris, H. Alan Fine, Gordon Harold Geiger, 2011-09-06 This book approaches the subject of material and energy balances from two directions First it emphasizes the fundamental principles of the conservation of mass and energy and the consequences of these two principles Second it applies the techniques of computational chemistry to materials processing and introduces new software developed by the author especially for material and heat balances The third edition reflects the changes in the professional engineer's practice in the last 30 years reflecting the dramatic shift away from metallurgical engineering and the extractive industry towards materials engineering A large and growing number of recent graduates are employed in such fields as semiconductor processing environmental engineering and the production and processing of advanced and exotic materials for aerospace electronic and structural applications The advance in computing power and software for the desktop computer has significantly changed the way engineers make computations and the biggest change comes from the computational approach used to solve problems The spreadsheet program Excel is used extensively throughout the text as the main computational engine for solving material and energy balance equations and for statistical analysis of data The use of Excel and the introduction of the add in programs enables the study of a range of variables on critical process parameters and emphasis is placed on multi device flowsheets with recycle bypass and purge streams whose material and heat balance equations were previously too complicated to solve by the normally used hand calculator The Excel based program FlowBal helps the user set up material and heat balance equations for processes with multiple streams and units **Holt Chemistry** Ralph Thomas Myers, 2004 **MCAT Comprehensive Review, 2004** Kaplan, Inc, 2003 The bestselling MCAT prep guide is a must for pre med students preparing for the demanding Medical College Admissions Test MCAT The CD features one full length practice test with detailed explanations plus effective tips and strategies **MCAT Comprehensive Review**, 2004 Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the

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