

STUDENT BOOK ANSWERS
To come
Chapter 9: Gas laws
Worked example 9.1

- 1 $n = \frac{V}{V_m} = \frac{2.33}{22.71} = 0.1026 \text{ mol}$
 $m = n \times MM = 0.1026 \times (12.01 + (4 \times 1.008)) = 1.646 = 1.65 \text{ g (3sf)}$
- 2 $n = \frac{m}{MM} = \frac{23.0}{14.01 + (3 \times 1.008)} = 1.350 \text{ mol}$
 $V = n \times V_m = 1.350 \times 24.79 = 33.47 = 33.5 \text{ L (3sf)}$

Worked example 9.2

- 1 $2\text{Na(s)} + 2\text{H}_2\text{O(l)} \rightarrow 2\text{NaOH(aq)} + \text{H}_2\text{(g)}$
 $n(\text{Na}) = \frac{m}{MM} = \frac{1.87}{22.99 + 16.00 + 1.008} = 0.04675 \text{ mol}$
 $n(\text{H}_2) = \frac{1}{2} \times n(\text{Na}) = 0.5 \times 0.04675 = 0.02338 \text{ mol}$
 $V = n \times V_m = 0.02338 \times 24.79 = 0.5795 = 0.580 \text{ L (3sf)}$
- 2 $n(\text{HCl}) = \frac{V}{V_m} = \frac{13.4}{22.71} = 0.5900 \text{ mol}$
 $2\text{KCl(aq)} + \text{H}_2\text{SO}_4\text{(aq)} \rightarrow 2\text{HCl(g)} + \text{K}_2\text{SO}_4\text{(aq)}$
 $n(\text{KCl}) = n(\text{HCl}) = 0.5900 \text{ mol}$
 $m(\text{KCl}) = n \times MM = 0.5900 \times (39.10 + 35.45) = 43.98 = 44.0 \text{ g (3sf)}$

Check your understanding 9.1/9.2

- 1 The volume of a gas changes with temperature and pressure, both of which compress or expand the gas particles.
- 2 Use the formula $n = \frac{V}{V_m}$ to convert the volume of gas to a number of moles. V_m depends upon the temperature of the gas. Use $m = n \times MM$ to convert the number of moles (n) to a mass (m) by multiplying by the molecular mass (MM).
- 3 **a** **i** $2 \times 250 = 500 \text{ mL}$
ii 250 mL
b **i** 500 mL
ii Nothing, as the law of combining volumes only applies to gases.

Chapter 9 Notes Cengage Learning

Jianjun Gao



Chapter 9 Notes Cengage Learning:

1, 2, 3 Code!-Instructor's Manual Ken Green,2006-08 Governance in Family Enterprises A. Koeberle-Schmid,D. Kenyon-Rouvinez,E. Poza,INOBUCHI,2013-12-02 Presents a comprehensive overview of governance in family enterprises including practical management knowledge in easy to use frameworks and interviews with renowned family enterprise owners and managers Readers will benefit from the book s systematic approach and the opportunity to learn from the experience of other family enterprises Science and Technology in World History, Volume 1 David Deming,2014-01-10 Science is a living organic activity the meaning and understanding of which have evolved incrementally over human history This book the first in a roughly chronological series explores the development of the methodology and major ideas of science in historical context from ancient times to the decline of classical civilizations around 300 A D It includes details specific to the histories of specialized sciences including astronomy medicine and physics along with Roman engineering and Greek philosophy It closely describes the contributions of such individuals as Pythagoras Hippocrates Socrates Plato Aristotle Alexander the Great Euclid Archimedes Ptolemy Seneca Pliny the Elder and Galen Science and Technology in World History, Volume 2 David Deming,2014-01-10 Science is a living organic activity the meaning and understanding of which have evolved incrementally over human history This book the second in a roughly chronological series explores the evolution of science from the advents of Christianity and Islam through the Middle Ages focusing especially on the historical relationship between science and religion Specific topics include technological innovations during the Middle Ages Islamic science the Crusades Gothic cathedrals and the founding of Western universities Close attention is given to such figures as Paul the Apostle Hippolytus Lactantius Cyril of Alexandria Hypatia Cosmas Indicopleustes and the Prophet Mohammed

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