

# Experiment 11A: The Periodic Table

## Procedure

### Part 1 Electronic Configuration and the Periodic Table

1. Watch the video presentation on page 1.

### Part 2 Atomic Number, Atomic Radius, Ionization Energy and the Periodic Table

1. Watch the video presentation on page 2.

### Part 3 Electronegativity and the Periodic Table

1. Watch the video presentation on page 3.

## Background

1. Examine the periodic table of elements and compare it to Figure 11.1. What relationship can be seen between an element's placement in the periodic table and its atomic configuration?

Elements in the periodic table are arranged in order of increasing atomic number and chemical properties. Though there are exceptions, most periodic trends are consistent.

2. Examine the graph of atomic radius plotted against increasing atomic number (Figure 11.4). What is particularly noteworthy about it? Is it similar to the periodic trend?

Atomic radius is the distance from the nucleus to the outermost electron shell. Elements in the periodic table show a general trend of increasing atomic number. The atomic radius increases from left to right and decreases from top to bottom. The atomic radius is the distance from the nucleus to the outermost electron shell.

3. Which group of elements has the smallest atomic radii? Which group has the largest?

Examine the graph of ionization energy plotted against increasing atomic number (Figure 11.5). What is particularly noteworthy about it? Is it similar to the periodic trend?

4. Examine the graph of electronegativity plotted against increasing atomic number (Figure 11.6). What is particularly noteworthy about it? Is it similar to the periodic trend?

Electronegativity is the ability of an atom to attract electrons. Elements in the periodic table show a general trend of increasing electronegativity from left to right and decreasing from top to bottom. The electronegativity is the ability of an atom to attract electrons.

# Laboratory Experiment 11a Answers Periodic Table

**JS Bruner**



**Laboratory Experiment 11a Answers Periodic Table:**

## Adopting the Melody of Appearance: An Psychological Symphony within **Laboratory Experiment 11a Answers Periodic Table**

In some sort of eaten by monitors and the ceaseless chatter of immediate communication, the melodic splendor and emotional symphony created by the prepared word often diminish in to the background, eclipsed by the constant sound and disturbances that permeate our lives. However, located within the pages of **Laboratory Experiment 11a Answers Periodic Table** a stunning fictional treasure overflowing with natural thoughts, lies an immersive symphony waiting to be embraced. Constructed by an outstanding composer of language, this captivating masterpiece conducts readers on a psychological trip, skillfully unraveling the concealed songs and profound influence resonating within each cautiously crafted phrase. Within the depths of this poignant evaluation, we shall explore the book is central harmonies, analyze its enthralling publishing style, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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