

## AP Chemistry Practice Test Chapter 8 and 9

### Multiple Choice

Identify the choice that best completes the statement or answers the question.

- \_\_\_\_\_ 1. Which pair of atoms should form the most polar bond?  
a. F and B  
b. C and O  
c. F and O  
d. N and F  
e. B and N
- \_\_\_\_\_ 2. Which pair of ions should form the ionic lattice with the highest energy?  
a.  $\text{Na}^{1+}$  and  $\text{Br}^{1-}$   
b.  $\text{Li}^{1+}$  and  $\text{F}^{1-}$   
c.  $\text{Cs}^{1+}$  and  $\text{F}^{1-}$   
d.  $\text{Li}^{1+}$  and  $\text{O}^{2-}$   
e.  $\text{K}^{1+}$  and  $\text{F}^{1-}$
- \_\_\_\_\_ 3. The electrostatic force of attraction is greatest in which compound?  
a.  $\text{BaO}$   
b.  $\text{MgO}$   
c.  $\text{CaS}$   
d.  $\text{MgS}$   
e.  $\text{CaO}$
- \_\_\_\_\_ 4. Which molecule has the weakest bond?  
a.  $\text{CO}$   
b.  $\text{O}_2$   
c.  $\text{NO}$   
d.  $\text{N}_2$   
e.  $\text{Cl}_2$
- \_\_\_\_\_ 5. For which species are octet resonance structures necessary and sufficient to describe the bonding satisfactorily?  
a.  $\text{BCl}_3$   
b.  $\text{SO}_2$   
c.  $\text{CO}_2$   
d.  $\text{BeF}_2$   
e.  $\text{SO}_4^{2-}$
- \_\_\_\_\_ 6. How are the bonding pairs arranged in the best Lewis structure for ozone?  
a.  $\text{O}-\text{O}-\text{O}$   
b.  $\text{O}=\text{O}-\text{O}$   
c.  $\text{O}=\text{O}=\text{O}$   
d.  $\text{O}=\text{O}=\text{O}$   
e.  $\text{O}=\text{O}=\text{O}$
- \_\_\_\_\_ 7. Which species has the shortest bond length?  
a.  $\text{CN}^{1-}$   
b.  $\text{O}_2$   
c.  $\text{SO}_2$   
d.  $\text{SO}_3$   
e.  $\text{CO}_2$
- \_\_\_\_\_ 8. Which species requires the least amount of energy to remove an electron from the outermost energy level?  
a.  $\text{Na}^{1+}$   
b. He  
c.  $\text{F}^{1-}$   
d.  $\text{O}^{2-}$   
e.  $\text{Mg}^{2+}$
- \_\_\_\_\_ 9. Which species has a valid non-octet Lewis structure?  
a.  $\text{GeCl}_4$   
b.  $\text{SiF}_4$   
c.  $\text{CO}_3^{2-}$   
d.  $\text{SeCl}_4$   
e.  $\text{CO}_3^{2-}$

# Chemistry Chapter 8 Test

**Thomas Evangelist**



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**Chemistry** Albert S. Tarendash, Frederick J. Rowe, 2002 This indispensable guide to chemistry helps students who wish to prepare for the AP Chemistry exam on their own Comprehensive and easy to understand this learning guide includes a full content review two full length practice tests with hundreds of practice questions and thorough answer explanations and proven test taking strategies

**The Therapeutic Gazette** William Brodie, Francis Edward Stewart, John J. Mulheron, Albert Brown Lyons, 1916

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