

CHAPTER 8

BONDING: GENERAL CONCEPTS

Questions

15. a. This diagram represents a polar covalent bond as in HCl. In a polar covalent bond, there is an electron rich region (indicated by the red color) and an electron poor region (indicated by the blue color). In HCl, the more electronegative Cl atom (on the red side of the diagram) has a slightly greater ability to attract the bonding electrons than does H (on the blue side of the diagram), which in turn produces a dipole moment.
- b. This diagram represents an ionic bond as in NaCl. Here, the electronegativity differences between the Na and Cl are so great that the valence electron of sodium is transferred to the chlorine atom. This results in the formation of a cation, an anion, and an ionic bond.
- c. This diagram represents a pure covalent bond as in H_2 . Both atoms attract the bonding

Chapter 8 Bonding General Concepts

Donald J. DeCoste



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Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement [Grade 10 Chemistry- simpleNeasyBook](#) WAGmob, 2014-03-27 WAGmob An eBook and app platform for learning teaching and training WAGmob brings you simpleNeasy on the go learning eBook for Grade 10 Chemistry The eBook provides 1 Snack sized chapters for easy learning This eBook provides a quick summary of essential concepts in Grade 10 Chemistry via easy to grasp snack sized chapters Atomic Structure Periodicity Bonding General Concepts Chemical Bonding Nuclear Reactions Thermodynamics Chemical Kinetics Chemical Equilibrium Gas and Gas laws Solid and Liquid About WAGmob eBooks 1 A companion eBook for on the go bite sized learning 2 Over Three million paying customers from 175 countries Why WAGmob eBooks 1 Beautifully simple Amazingly easy Massive selection of eBook 2 Effective Engaging and Entertaining eBook 3 An incredible value for money Lifetime of free updates WAGmob Vision simpleNeasy eBooks for a lifetime of on the go learning WAGmob Mission A simpleNeasy WAGmob eBook in every hand WAGmob Platform A unique platform to create and publish your own apps e Books Please visit us at www.wagmob.com or write to us at Team.wagmob.com We would love to improve our eBook and app platform [Wong's Nursing Care of Infants and Children Multimedia Enhanced Version](#) Marilyn J. Hockenberry, David Wilson, 2013-10-17 A most trusted authority in pediatric nursing this book provides unmatched comprehensive coverage of pediatric growth development and conditions Its unique age and stage approach covers child development and health promotion as well as specific health problems organised by age groups and body systems **Chemistry** Steven S. Zumdahl, Kenneth C. Brooks, 1986 **Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl** Paul B. Kelter, 1993 **Study Guide for Chemistry by Steven S. Zumdahl** Martha B. Barrett, 1986 **Solutions Guide for Chemistry, Third Edition, by Steven S. Zumdahl** Kenneth C. Brooks, 1993 **Active Learning Guide for Chemistry** Donald J. DeCoste, 2007 **Beyond the Visible Universe : from a New Space-time Concept of the Physical Vacuum** S. Sarg, 2005 **Solutions Guide for Chemistry** Steven S. Zumdahl, Kenneth C. Brooks, 1989 **Basic Chemical Concepts and Tables** Steven L. Hoenig, 2024-05-16 Fully revised and expanded the second edition of Basic Chemical Concepts and Tables is written as a quick reference to the many different concepts and ideas encountered in chemistry The volume presents important subjects in a concise format that makes it a

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