

62 Test - The Chemistry of Life

BIO 1A

- ___ 1. All matter is composed of. a) atoms b) cells c) carbon d) hydrogen
- ___ 2. The periodic table displays _____ arranged by their atomic number.
a) molecules b) protons c) elements d) neutrons
- ___ 3. H₂O is an example of a(n): a) element b) subatomic particle c) atom d) compound
- ___ 4. Which particles are found within the nucleus of an atom?
a) protons and electrons b) protons and neutrons
c) electrons and neutrons d) protons, electrons, and neutrons
- ___ 5. A gecko's feet stick to surfaces due to:
a) cohesion b) adhesion c) hydrogen bonds d) van der Waals forces
- ___ 6. What is responsible for the "stickiness" of water that results in cohesion and adhesion?
a) covalent bonds b) ionic bonds c) hydrogen bonds d) van der Waals forces
- ___ 7. Polar molecules, like water, have what characteristic?
a) partial negative and positive charges
b) multiple hydrogen bonds
c) an uneven number of protons and electrons
d) long chains of carbon
- ___ 8. A water strider can move across the surface of water due to what property of water?
a) adhesion b) cohesion c) polarity d) heat capacity
- ___ 9. Most enzymes found in the body work best at about 40 degrees. This represents the enzyme's:
a) base rate b) substrate c) optimal pH d) optimal temperature
- ___ 10. What is pH?
a) a measurement of how acidic or basic a substance is
b) the optimal temperature enzymes operate at
c) the tendency of water to attract other water molecules
d) the tendency of water to resist changes in state
- ___ 11. Substances that speed up chemical reactions in the body are called:
a) substrates b) enzymes c) carbohydrates d) ions
- ___ 12. What type of bond results from the sharing of electrons?
a) ionic b) carbonated c) covalent d) van der Waals

Chapter 2 The Chemistry Of Life Test B Answer Key

William K. Purves



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