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19.8 (Pages 871 - 877) ; Watch [VSE 19.8](#)

Consider the reaction



Estimate ΔG° for this reaction at each temperature. (Assume that ΔH° and ΔS° do not change too much within the given temperature range.)

Substance	ΔH_f° (kJ/mol)	S° (J/(mol · K))
NO(g)	91.3	210.8
O ₂ (g)	0	205.2
NO ₂	33.2	240.1

Part A

298 K

Express your answer in kilojoules to three significant figures.

□
√
Δ
Σ
↶
↷
⊙
⊞
?

$\Delta G^\circ =$ kJ

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Part B

Pages 871 876 Answers

Michael Seilmaier



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