

2023-24

**KVS PGT**  
**Computer Science**

**Youth**  
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**Solved Papers & Practice Book**

**Kendriya Vidyalaya Sangathan**

# KVS PGT

**Post Graduate Teacher**

# Computer Science

**Solved Papers &  
Practice Book**

**15  
Sets**

### UP-TO-DATE SYLLABUS

Sl. No.	Subject	Marks
Part-1	General English	15
	General Hindi	15
	General Knowledge & Current Affairs	15
Part-2	Reasoning	15
	Computer Literacy	15
Part-3	Proficiency on Education & Leadership	15
Part-4	Computer Subject	150

**Total No. of Questions: 100, Marks: 180, Time: 120 Minutes**

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based on REVISED  
ANSWER KEY**

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## **Kvs Pgt Computer Science Question Papers:**

### **PGT Computer Science Previous Year Solved MCQs Bilingual Edition KVS/NVS/DSSSB/State PGT's/Others**

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web les principaux genres d orchidées vendus actuellement sont cattleya cymbidium dendrobium oncidium odontoglossum paphiopedilum phalaenopsis miltoniopsis vanda vanilla et zygopetalum elles fascinent par l étrange beauté de leurs fleurs depuis des siècles les passionnés les collectionnent avec ferveur focus sur les orchidées

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web may 31 2010 les orchidées sont de belles fleurs exotiques qui pour se reproduire utilisent d incroyables artifices c est sans doute ce qui les rend si fascinantes

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web jul 13 2022 partager sur on vous a peut être déjà offert un pot de phalaenopsis c est en effet l une des principales orchidées vendues en jardinerie et chez les fleuristes mais la famille des orchidées regroupe bien d autres espèces les scientifiques en recensent plus de 25 000 au total d apparences infiniment variées

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web publié le 25 juillet 2020 les orchidées sont dans le règne végétal la famille de plantes qui serait la plus évoluée au niveau de la florescence de la reproduction et de ses liens

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web l orchidée est l une des fleurs les plus appréciées mais quelle est sa signification dans le langage des fleurs quand offrir une orchidée en cadeau et quelles sont les légendes les plus fascinantes qui la considèrent comme le protagoniste

*lab report copper iron stoichiometry 1522 words cram* - Jul 02 2022

web the purpose of the copper iron stoichiometry lab was to determine which of the two possible iron sulfate compounds formed when reacting aqueous copper ii sulfate with solid iron metal the techniques used in this lab were weighing by difference quantitative transfer and vacuum filtration

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the ion meta dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab

questions write the balanced equation

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate  $\text{CuSO}_4$  and 2 0153 grams of iron powder  $\text{Fe}$  the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

**lab 4 lab 4 stoichiometry of iron copper ii sulfate redox** - Jan 08 2023

web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web stoichiometric principles can be used to determine the chemical equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction the iron ions dissolve in the copper ii sulfate solution to produce copper ii ions

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web lab 4 stoichiometry the reaction of iron with copper ii sulfate revised 8 19 2009 1 introduction in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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web stoichiometry lab report the reaction of iron with copper ii sulfate period 5 in this experiment i observed the reaction of iron with copper ii sulfate i obtained the results to this experiment through adding heated  $\text{CuSO}_4$  solution to the iron powder stirring the solution adding water then removing water adding acetone heating the acetone on low

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reactant in a chemical reaction

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web stoichiometry lab iron and copper sulfate sample data data and observations mass of copper ii sulfate pentahydrate 3 27  
g of g mass of filter paper mass of filter paper and dry copper mass of dry copper 1 28 g