

## Review Problems &amp; Solutions

The test is 'closed book,' but if you wish you may bring a one-sided sheet of formulas. The intent of this sheet is as a reminder of important formulas and definitions, and as a compact transcription of the answers provided here. If this privilege is abused, it will be revoked for future tests. The test will be composed entirely from a subset of the following problems, as well as those in problem sets 3 and 4. Thus if you are familiar and comfortable with these problems, there will be no surprises!

\*\*\*\*\*

1. *Scaling in fluids:* Near the liquid-gas critical point, the free energy is assumed to take the scaling form  $f/N = t^{2-\alpha} g(\delta\rho/t^\beta)$ , where  $t = |T - T_c|/T_c$  is the reduced temperature, and  $\delta\rho = \rho - \rho_c$  measures deviations from the critical point density. The leading singular behavior of any thermodynamic quantity  $Q(t, \delta\rho)$  is of the form  $t^{\nu_Q}$  on approaching the critical point along the isochore  $\rho = \rho_c$ ;  $\propto \delta\rho^\beta$  for a path along the isotherm  $T = T_c$ . Find the exponents  $\alpha$  and  $\beta$  for the following quantities:

- Any homogeneous thermodynamic quantity  $Q(t, \delta\rho)$  can be written in the scaling form

$$Q(t, \delta\rho) = t^{\nu_Q} g_Q\left(\frac{\delta\rho}{t^\beta}\right).$$

Thus, the leading singular behavior of  $Q$  is of the form  $t^{\nu_Q}$  if  $\delta\rho = 0$ , i.e. along the critical isochore. In order for  $Q$  to be independent of  $t$  along the critical isotherm  $t = 0$ , the scaling function for a large enough argument should be of the form

$$\lim_{x \rightarrow \infty} g_Q(x) = x^{-\nu_Q/\beta},$$

so that

$$Q(0, \delta\rho) \propto (\delta\rho)^{\nu_Q}, \quad \text{with} \quad \nu_Q = \frac{\nu_Q}{\beta}.$$

(a) The internal energy per particle  $U/N$ , and the entropy per particle  $S/N$ .

- Let us assume that the free energy per particle is

$$f = \frac{F}{N} = t^{2-\alpha} g\left(\frac{\delta\rho}{t^\beta}\right),$$

and that  $T < T_c$ , so that  $\frac{\partial}{\partial T} = -\frac{1}{T} \frac{\partial}{\partial t}$ . The entropy is then given by

$$s = -\left.\frac{\partial f}{\partial T}\right|_V = \frac{1}{T_c} \left.\frac{\partial f}{\partial t}\right|_\rho = \frac{t^{1-\alpha}}{T_c} g_s\left(\frac{\delta\rho}{t^\beta}\right),$$

# Statistical Mechanics Solution

**Jiyuan Zhang**



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