

Name: \_\_\_\_\_

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Chemistry 11

**Classifying Matter**

Assignment

1. Complete the following sentences by filling in the appropriate word from the list below.

Elements  
Mixtures  
Compound  
Symbols  
Properties

Carbon  
Pure  
Letters  
Carbon dioxide

- a. \_\_\_\_\_ **Elements** \_\_\_\_\_ cannot be separated by physical or chemical means.
- b. Elements are abbreviated with \_\_\_\_\_ **symbols** \_\_\_\_\_, which consist of one or two \_\_\_\_\_ **letters** \_\_\_\_\_.
- c. Two or more elements chemically combined make up a \_\_\_\_\_ **compound** \_\_\_\_\_.
- d. \_\_\_\_\_ **Carbon** \_\_\_\_\_ is an example of an element and \_\_\_\_\_ **carbon dioxide** \_\_\_\_\_ is an example of a compound.
- e. Elements and compounds are called \_\_\_\_\_ **pure** \_\_\_\_\_ substances because they have a unique set of chemical and physical \_\_\_\_\_ **properties** \_\_\_\_\_.

2. Fill in the blanks in the following chart.

Element Name	Element Symbol	Common Ions Formed
a. Sodium	<b>Na</b>	<b>+1</b>
b. Copper	Cu	<b>+2, +1</b>
c. Tungsten	W	<b>+6</b>
d. Iron	<b>Fe</b>	<b>+3, +2</b>
e. Tin	Sn	<b>+4, +2</b>
f. Potassium	<b>K</b>	<b>+1</b>
g. Gold	Au	<b>+3, +1</b>
h. Silver	Ag	<b>+1</b>

3. Classify each of the following as one of an Atom (A), a Molecule (M) or an Ion (I).

a.  $P^{3-}$       I    
 b.  $H_2O$       M    
 c.  $O_2$       M    
 d.  $Au^{3+}$       I  

e. Ge      A    
 f. O      A    
 g.  $Ca^{2+}$       I    
 h.  $NH_3$       M

# Chemistry A Study Of Matter 114 Answer Key

**N Colangelo**



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**The University of Idaho Bulletin** University of Idaho,1975 *Catalogue of the Officers and Students of Antioch College of Yellow Springs, Greene Co., Ohio, for the Academical Year* Antioch College,1929 **Instructor's Manual and Study Guide Answers for the Human Body in Health and Disease** Barbara Janson Cohen,Memmler,1996 **Urban Airborne Particulate Matter** Fathi Zereini,Clare L. S. Wiseman,2011-02-16 This book presents the most up to date research and information regarding the origin chemistry fate and health impacts of airborne particulate matter in urban areas a topic which has received a great deal of attention in recent years due to documented relationships between exposure and health effects such as asthma With internationally recognised researchers and academics presenting their work and key concepts and approaches from a variety of disciplines including environmental and analytical chemistry biology toxicology mineralogy and the geosciences this book addresses the topic of urban airborne particulate matter in a comprehensive multidisciplinary manner Topics and research addressed in the book range from common methodological approaches used to sample and analyse the composition of airborne particulates to our knowledge regarding their potential to impact human health and the various policy approaches taken internationally to regulate particulate matter levels **British Books in Print** ,1979 **The Engineer** ,1919 **English Mechanic and World of Science** ,1879 **University of Virginia** Paul Brandon Barringer,James Mercer Garnett,Rosewell Page,1904 [The Naturalists' Leisure Hour and Monthly Bulletin](#) ,1892 **EPA Publications Bibliography** United States. Environmental Protection Agency,1996 **EPA Publications Bibliography Quarterly Abstract Bulletin** United States. Environmental Protection Agency,1996-07 **English Mechanic and Mirror of Science and Art** ,1880 **Index Medicus** ,2002 Vols for 1963 include as pt 2 of the Jan issue Medical subject headings *The Literary Digest* ,1921 [Holt Chemistry](#) Ralph Thomas Myers,2004 *English Mechanic and Mirror of Science* ,1873 **Journal of the Society of Chemical Industry** Society of Chemical Industry (Great Britain),1923 **Sir Thomas Browne** Reid Barbour,2013-08 Reid Barbour brings the historical evidence of Browne s life together for the first time allowing readers to contextualise his most celebrated works **Encyclopedia of Interfacial Chemistry** ,2018-03-29 Encyclopedia of Interfacial Chemistry Surface Science and Electrochemistry Seven Volume Set summarizes current fundamental knowledge of interfacial chemistry bringing readers the latest developments in the field As the chemical and physical properties and processes at solid and liquid interfaces are the scientific basis of so many technologies which enhance our lives and create new opportunities its important to highlight how these technologies enable the design and optimization of functional materials for heterogeneous and electro catalysts in food production pollution control energy conversion and storage medical applications requiring biocompatibility drug delivery and more This book provides an interdisciplinary view that lies at the intersection of these fields Presents fundamental knowledge of interfacial chemistry surface science and electrochemistry and provides cutting edge research from academics and practitioners across

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