

Gravimetric Analysis Questions

Q-1: Which of the following crucibles is used when precipitates are dried in the muffle furnace in gravimetric analysis?

- a) Crucible made of porcelain
- b) Crucible made of silica
- c) Both of the above
- d) None of the above

Answer: c) Both of the above

Explanation: Crucible made of porcelain and silica are used when precipitates are dried in the muffle furnace in gravimetric analysis.

Q-2: Which of the following gravimetric analysis uses temperature as a function to measure a material's physical and chemical properties?

- a) Electrogravimetry
- b) Thermogravimetry
- c) Volatilisation gravimetry
- d) Precipitation gravimetry

Answer: b) Thermogravimetry is a thermal analysis method that measures changes in physical and chemical properties of materials as a function of increasing temperature or time.

Q-3: What is the principle of gravimetric analysis?

Answer: The gravimetric analysis principle is based on determining the mass % of an ion in a known amount of impure compound. It is then used to determine the mass % of the same ion in a known amount of impure substance.

Q-4: What are the three theories that govern gravimetry?

Answer: The law of mass action and reversible reactions, the principle of solubility product, and the common ion effect are the fundamental principles and theories of gravimetric analysis.

Q-5: What is the distinction between gravimetric and volumetric analysis?

Answer: Gravimetric and volumetric analysis are both quantitative methods for calculating the amount of sample in a solution or the purity of a compound; gravity analysis determines the mass of the analyte, while volumetric analysis determines the volume of the analyte.

Q-6: What do you mean by precipitation gravimetry?

Answer: Precipitation gravimetry is an analytical technique that uses a precipitation reaction to separate the ions in a solution. The precipitant, also known as the precipitating agent, is the chemical

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