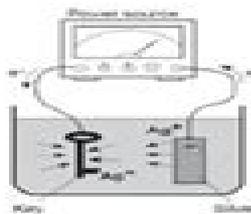


ENHANCING TEST ITEMS – CHEMISTRY (for Physical Sciences majors)

- Which of the following statements best describes the atomic radius of potassium atom and the atomic radius of a calcium atom?
 - The radius of the potassium atom is smaller because of its smaller nuclear charge.
 - The radius of the potassium atom is smaller because of its larger nuclear size.
 - The radius of the potassium atom is larger because its nuclear charge is smaller.
 - The radius of the potassium atom is larger because its nuclear charge is also larger.
 - Which type of chemical reaction occurs when an iron nail rusts?
 - condensation
 - neutralization
 - oxidation-reduction
 - ionization-dissociation
 - Which electron dot symbol represents the atom in period-4 with the highest first ionization energy?
 - 
 - 
 - 
 - 
 - Which phase change is exothermic?
 - $\text{H}_2\text{O}_{(l)} \rightarrow \text{H}_2\text{O}_{(g)}$
 - $\text{H}_2\text{O}_{(g)} \rightarrow \text{H}_2\text{O}_{(l)}$
 - $\text{H}_2\text{O}_{(l)} \rightarrow \text{H}_2\text{O}_{(s)}$
 - $\text{H}_2\text{O}_{(s)} \rightarrow \text{H}_2\text{O}_{(l)}$
 - Given the diagram, which statement BEST describes the key?
 - It is negative and acts as the cathode.
 - It is positive and acts as the cathode.
 - It is negative and acts as the anode.
 - It is positive and act as the anode
- 
- Which technique is safest for diluting a concentrated acid with water?
 - Add the acid to the water quickly.
 - Add the water to the acid quickly.
 - Add the acid to the water slowly while stirring constantly.
 - Add the water to the acid slowly while stirring constantly.
 - Which elements combine to form an ionic bond?
 - sodium and potassium
 - sodium and oxygen
 - carbon and oxygen
 - carbon and sulfur
 - As the elements in group 16 are considered in order from top to bottom, the atomic radius of each successive element increases. This increase is primarily due to an increase in
 - electronegativity
 - ionization energy
 - the number of protons occupying the nucleus
 - the number of occupied principal energy levels
 - An air pollution situation in which reactions occur on the surfaces of solid particles and produce acidic irritation of lung tissues is called _____.

Chemistry Practice Test With Answers

Kaplan



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Getting into Medical School For Dummies Carleen Eaton,2013-05-03 Your plain English guide to getting into the medical school of your dreams

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AP Chemistry For Dummies Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13. A practical and hands-on guide for learning the practical science of AP chemistry and preparing for the AP chem exam. Gearing up for the AP Chemistry exam, *AP Chemistry For Dummies* is packed with all the resources and help you need to do your very best. Focused on the chemistry concepts and problems the College Board wants you to know, this AP Chemistry study guide gives you winning test-taking tips, multiple-choice strategies, and topic guidelines, as well as great advice on optimizing your study time and hitting the top of your game on test day. This user-friendly guide helps you prepare without perspiration by developing a pre-test plan, organizing your study time, and getting the most out of your AP course. You'll get help understanding atomic structure and bonding, grasping atomic geometry, understanding how colliding particles produce states, and so much more. To provide students with hands-on experience, AP chemistry courses include extensive labwork as part of the standard curriculum. This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques, and another to a complete survey of recommended AP chemistry experiments. Two full-length practice exams help you build your confidence, get comfortable with test formats, identify your strengths and weaknesses, and focus your studies. You'll discover how to create and follow a pretest plan, understand everything you must know about the exam, develop a multiple-choice strategy, figure out displacement, combustion, and acid-base reactions, get familiar with stoichiometry, describe patterns, and predict properties. Get a handle on organic chemistry nomenclature, know your way around laboratory concepts, tasks, equipment, and safety, analyze laboratory data, use practice exams to maximize your score. Additionally, you'll have a chance to brush up on the math skills that will help you on the exam, learn the critical types of chemistry problems, and become familiar with the annoying exceptions to chemistry rules. Get your own copy of *AP Chemistry For Dummies* to build your confidence and test-taking know-how so you can ace that exam.

CRACKING THE AP ENVIRONMENTAL SCIENCE EXAM (2011 EDITION) Princeton Review, 2011-03-08. Reviews topics covered on the test, offers tips on test-taking strategies, and includes two full-length practice tests with answers and explanations.

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AP Chemistry Albert S. Tarendash, Frederick J. Rowe, 2002 This indispensable guide to chemistry helps students who wish to prepare for the AP Chemistry exam on their own Comprehensive and easy to understand this learning guide includes a full content review two full length practice tests with hundreds of practice questions and thorough answer explanations and proven test taking strategies

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