

Chemical Reactions and Equations

Chemical Reaction

- change in state
- colour
- gas
- temperature

ex: Magnesium + Oxygen \rightarrow

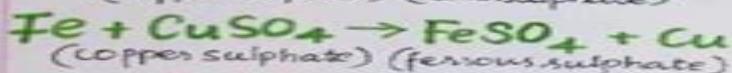
(Reactants)

magnesium oxide
(products)



DISPLACEMENT

Reaction

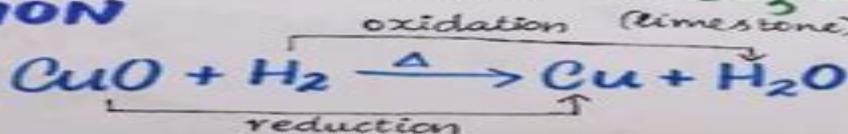


DOUBLE DISPLACEMENT

Reaction



OXIDATION & REDUCTION



Why balance Chemical equation

\Rightarrow Mass can neither be created nor destroyed in a chemical reaction.
 \triangleright Total mass of elements in the product of a chemical reaction has to be equal to the total mass of the elements in the reactants side.

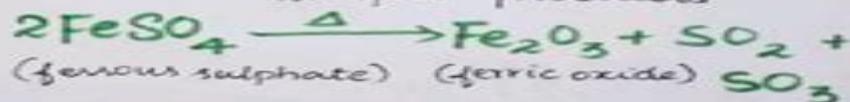
COMBINATION

Reaction when two or more substances combine to form a single product.



DECOMPOSITION

Reaction when a single reactant breaks down to simpler products



Chemistry Notes Chapter 8 Chemical Reactions

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