

$n_1 \rightarrow$  Lyman (1) (UV Region)  
 $n_2 \rightarrow$  Balmer (2) (visible)  
 $n_3 \rightarrow$  3 Paschen  
 $n_4 \rightarrow$  4 Brackett  
 $n_5 \rightarrow$  5 p-fund.

Time period =  $\frac{2\pi r}{v}$   $\propto \frac{n^3}{Z^2}$

Frequency =  $\frac{1}{T}$   $\propto \frac{Z^2}{n^3}$

No. of spectral lines =  $\frac{(n_2 - n_1)(n_2 - n_1 + 1)}{2}$

$\Delta E \propto \frac{1}{n^2}$

No. of spectral lines if  $n=1 = \frac{n(n-1)}{2}$

No. of spectral lines.  
 Lyman =  $n-1$ , Balmer =  $n-2$   
 Paschen =  $n-3$ , Brackett =  $n-4$   
 p-fund =  $n-5$

Photoelectric effect.  
 $E_{\text{photon}} = h\nu_0 + KE$   
 $h\nu_p = h\nu_0 + KE$

Stopping potential  $\Rightarrow E_{\text{photon}} - W_0 = qV$

Energy of  $e^- \rightarrow n+1$

de-Broglie's Eq.

$2\pi r = n\lambda$

No. of waves made in  $n^{\text{th}}$  orbit =  $n$

$\lambda = \frac{h}{p} \Rightarrow \lambda = \frac{h}{\sqrt{2m \cdot KE}} \quad \lambda = \frac{h}{\sqrt{2m \cdot qV}} \quad \lambda_e = \sqrt{\frac{150}{V}} \text{ \AA} = \frac{12.3}{\sqrt{V}} \text{ \AA}$

$\Delta x \Delta p \geq \frac{h}{4\pi} \quad \Delta n \cdot m \Delta v \geq \frac{h}{4\pi} \quad \Delta \theta \Delta \phi \geq \frac{h}{4\pi}$ 
  
 $\theta \rightarrow$  angular disp.  $\phi \rightarrow$  momentum.

No. of subshells in  $n^{\text{th}}$  shell or no. of orbitals  $l$  values from = 0 to  $(n-1)$

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