

Chapter Test II

A. Matching

Match each item in Column I with the correct description in Column II. Write the letter of the correct description.

Column I

1. An oxidation-reduction reaction in which an element is oxidized and another element is reduced.
2. An oxidation-reduction reaction in which a substance is oxidized and the same substance is reduced.
3. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.
4. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.
5. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.
6. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.

Column II

- a. redox reaction
- b. disproportionation
- c. combustion reaction
- d. redox reaction
- e. redox reaction
- f. redox reaction

B. Multiple Choice

Choose the best answer for each question.

1. Which of the following is a redox reaction?
 - a. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - b. $\text{NaCl} + \text{KBr} \rightarrow \text{NaBr} + \text{KCl}$
 - c. $\text{H}_2\text{O} + \text{CO}_2 \rightarrow \text{H}_2\text{CO}_3$
 - d. $\text{Ca} + \text{Mg} \rightarrow \text{CaMg}$
2. Which of the following is a redox reaction?
 - a. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$
 - b. $\text{NaCl} + \text{KBr} \rightarrow \text{NaBr} + \text{KCl}$
 - c. $\text{H}_2\text{O} + \text{CO}_2 \rightarrow \text{H}_2\text{CO}_3$
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Chemistry Chapter 12 Stoichiometry Test

Richard H. Langley, John Moore



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