

## Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 3

**Q1.** In a hydrocarbon, the mass ratio of hydrogen and carbon is 1:3; the empirical formula of hydrocarbon is \_\_\_\_\_.

- a.)  $\text{CH}_4$
- b.)  $\text{CH}_2$
- c.)  $\text{C}_2\text{H}$
- d.)  $\text{CH}_3$

**Q2.** 80g of oxygen contains as many atoms as in \_\_\_\_\_.

- a.) 10 g of hydrogen
- b.) 5 g of hydrogen
- c.) 8 g of hydrogen
- d.) 1 g of hydrogen

**Q3.** A mixture of gases contains  $\text{H}_2$  and  $\text{O}_2$  gases in the ratio 1:4. What is the molar ratio of the two gases in the mixture?

- a.) 16:1
- b.) 2:1
- c.) 1:4
- d.) 4:1

**Q4.** Which of the following concentration terms is/are independent of temperature?

- a.) Molality
- b.) Mole fraction
- c.) Molarity
- d.) Normality

**Q5.** Which of the following contains the highest number of atoms?

- a.) 1.0 g of water
- b.) 1.0 g of silver
- c.) 1.0 g of nitrogen
- d.) 1.0 g of propane

# Chemistry Chapter 17 Assessment Answers

**JE Gale**



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