

### Honors Chemistry - Chapter 12 Test

#### True/False - Clearly indicate the correct choice

1. In an aqueous solution of table salt (NaCl), the table salt would be considered the solvent.  
False
2. Water and gasoline are two liquids that would be considered miscible.  
False
3. A negative sign listed for the enthalpy of solution means that energy is released in the formation of the solution.  
True
4. A chemist finds that increasing the temperature leads to less of a substance dissolving in water. The substance is most likely a gas.  
True
5. Salt or sugar dissolved in water is considered to be a homogeneous mixture.  
True

#### Matching Set - Clearly indicate the correct choice

- a. solution    b. colloid    c. suspension
6. largest solute particle size c
  7. cannot be separated by filtration, will not scatter light a
  8. heterogeneous, does not separate on standing c
  9. particles settle out a
  10. smallest solute particle size b
  11. muddy water c
  12. shaving cream b

#### Multiple Choice - Clearly indicate the correct choice

13. A solute dissolved in alcohol is known as a/an  
a. aqueous solution    b. heterogeneous mixture    c. tincture  
d. immiscible solution  
c
14. An electrolyte is a substance that dissolves in water to form a solution

# Chemistry Chapter 12 Test Answers

**Wolfgang Guggemos**



**Chemistry Chapter 12 Test Answers:**

## **Chemistry Chapter 12 Test Answers** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has are more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such could be the essence of the book **Chemistry Chapter 12 Test Answers**, a literary masterpiece that delves deep to the significance of words and their affect our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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